



SUBATOMIC STRUCTURE

MAJOR SUBATOMIC PARTICLES

**MAJOR
SUBATOMIC PARTICLES**

PROTONS

**MAJOR
SUBATOMIC PARTICLES**

**MAJOR
SUBATOMIC PARTICLES**

**PROTONS
NEUTRONS**

**MAJOR
SUBATOMIC PARTICLES**

MAJOR SUBATOMIC PARTICLES

PROTONS

NEUTRONS

ELECTRONS

MAJOR SUBATOMIC PARTICLES



PROTONS

**MAJOR
SUBATOMIC PARTICLE
PROTONS**



POSITIVE CHARGE (+)

**MAJOR
SUBATOMIC PARTICLE
PROTONS**

**MAJOR
SUBATOMIC PARTICLE
PROTONS**

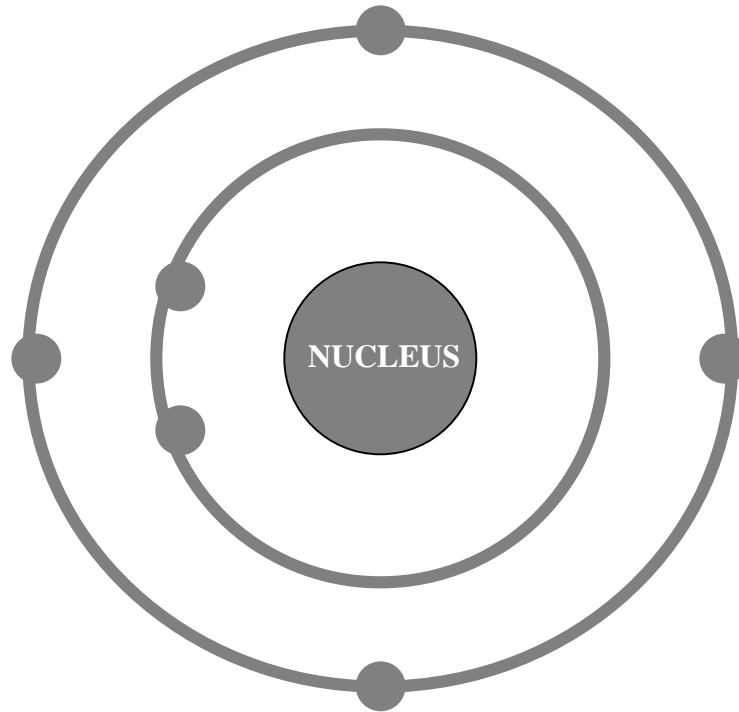


POSITIVE CHARGE (+)

WITHIN NUCLEUS

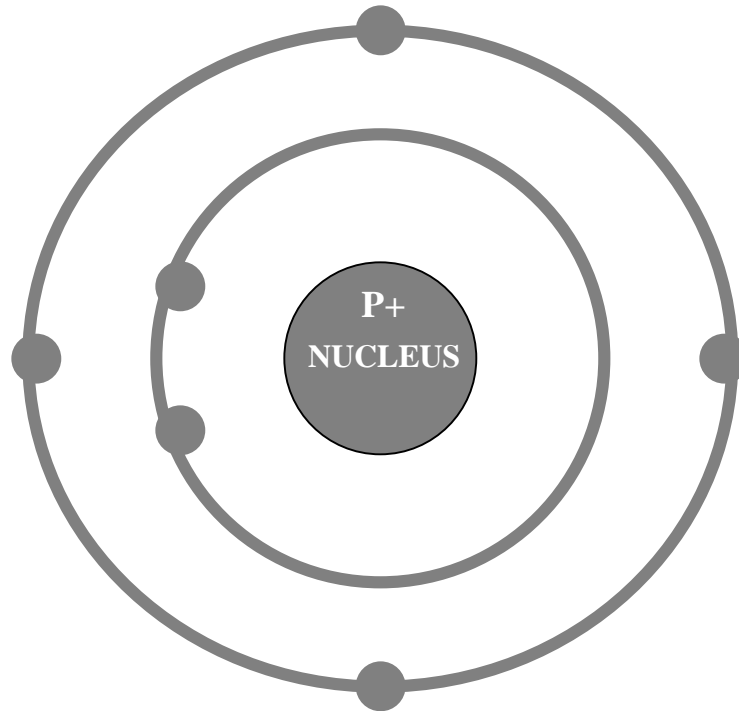
**MAJOR
SUBATOMIC PARTICLE
PROTONS**

SUBATOMIC PARTICLES



TYPICAL ATOM

SUBATOMIC PARTICLES



TYPICAL ATOM
PROTONS WITHIN NUCLEUS



NEUTRONS

**MAJOR
SUBATOMIC PARTICLE
NEUTRONS**



NEUTRAL CHARGE (~)

**MAJOR
SUBATOMIC PARTICLE
NEUTRONS**



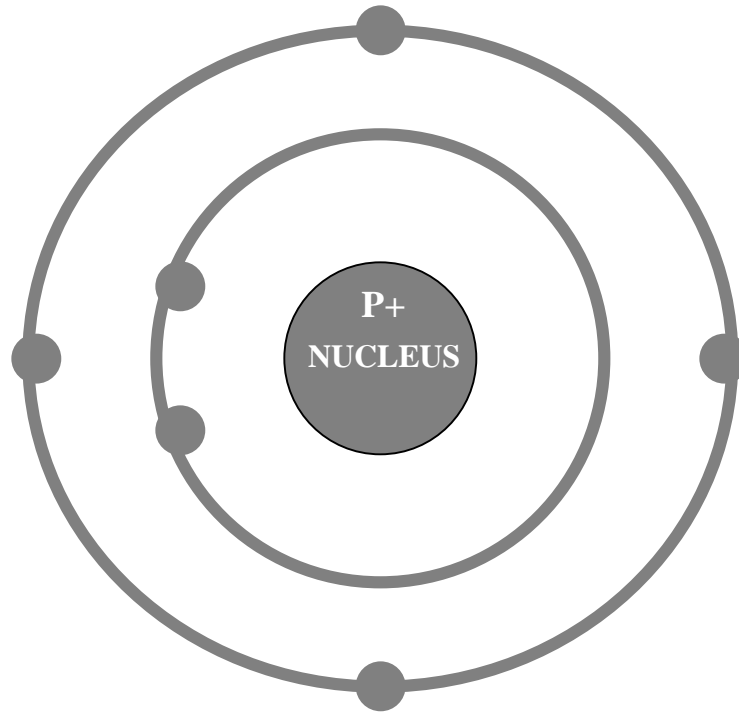
**MAJOR
SUBATOMIC PARTICLE
NEUTRONS**

NEUTRAL CHARGE (~)

WITHIN NUCLEUS

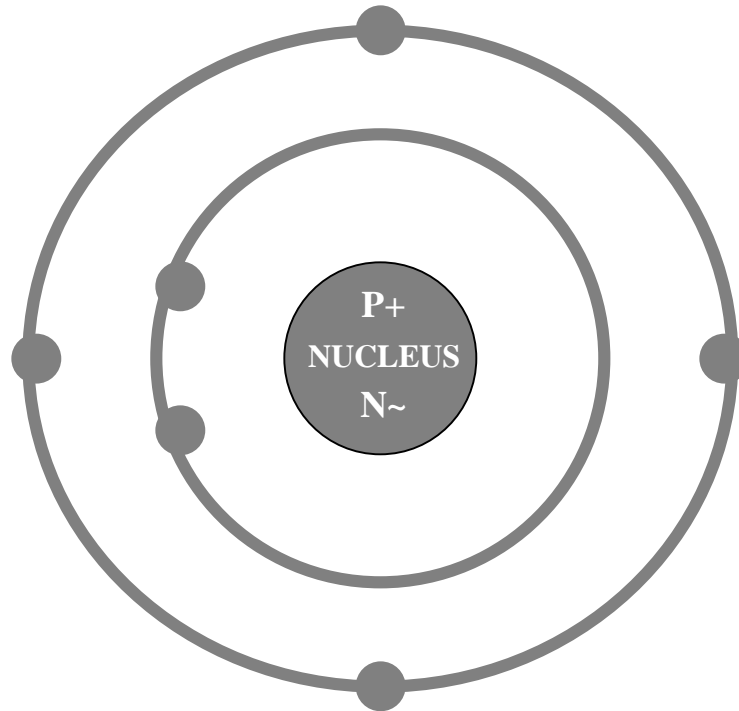
**MAJOR
SUBATOMIC PARTICLE
NEUTRONS**

SUBATOMIC PARTICLES



TYPICAL ATOM

SUBATOMIC PARTICLES



TYPICAL ATOM
NEUTRONS WITHIN NUCLEUS



ELECTRONS

**MAJOR
SUBATOMIC PARTICLE
ELECTRONS**



NEGATIVE CHARGE (-)

**MAJOR
SUBATOMIC PARTICLE
ELECTRONS**



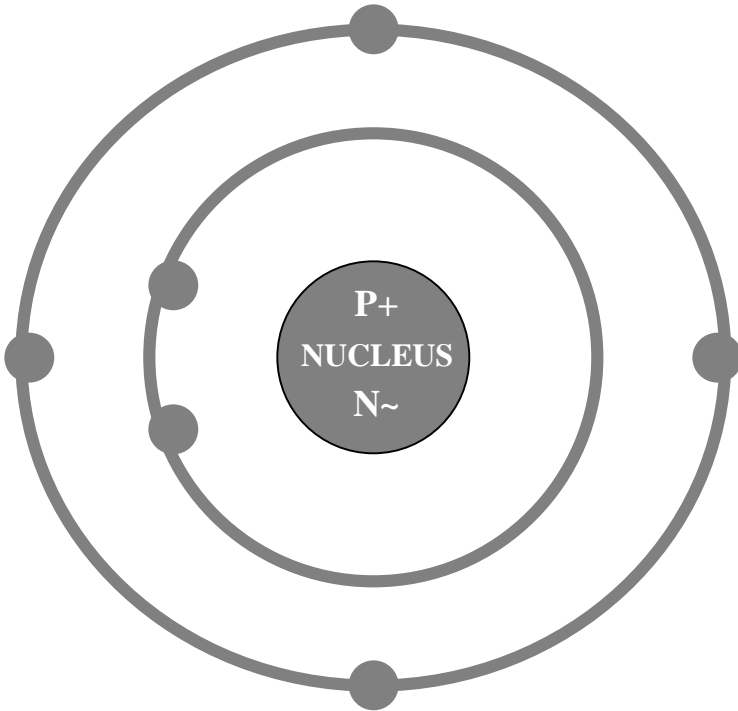
**MAJOR
SUBATOMIC PARTICLE
ELECTRONS**

NEGATIVE CHARGE (-)

ORBITING NUCLEUS

**MAJOR
SUBATOMIC PARTICLE
ELECTRONS**

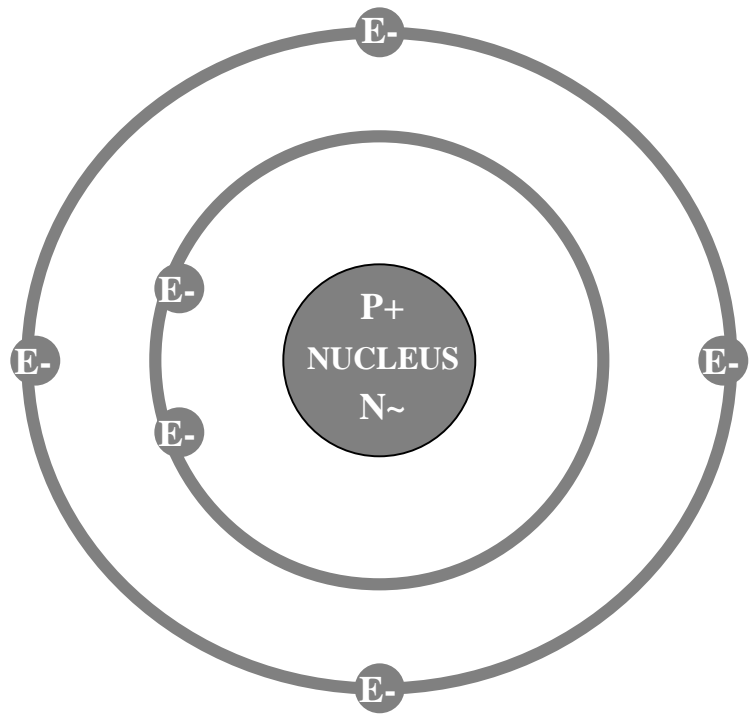
SUBATOMIC PARTICLES



TYPICAL ATOM



SUBATOMIC PARTICLES

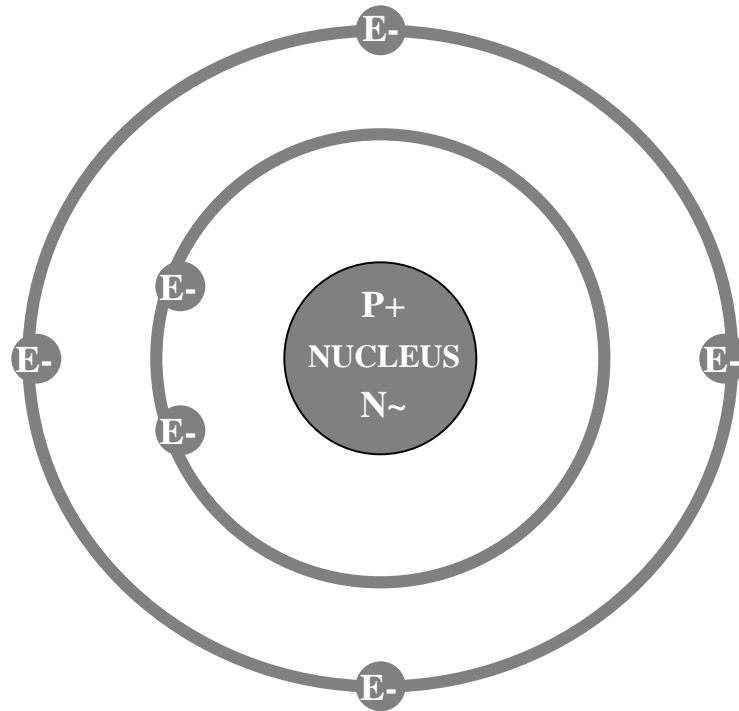


TYPICAL ATOM
ELECTRONS ORBIT NUCLEUS



MAJOR SUBATOMIC PARTICLES APPLIED

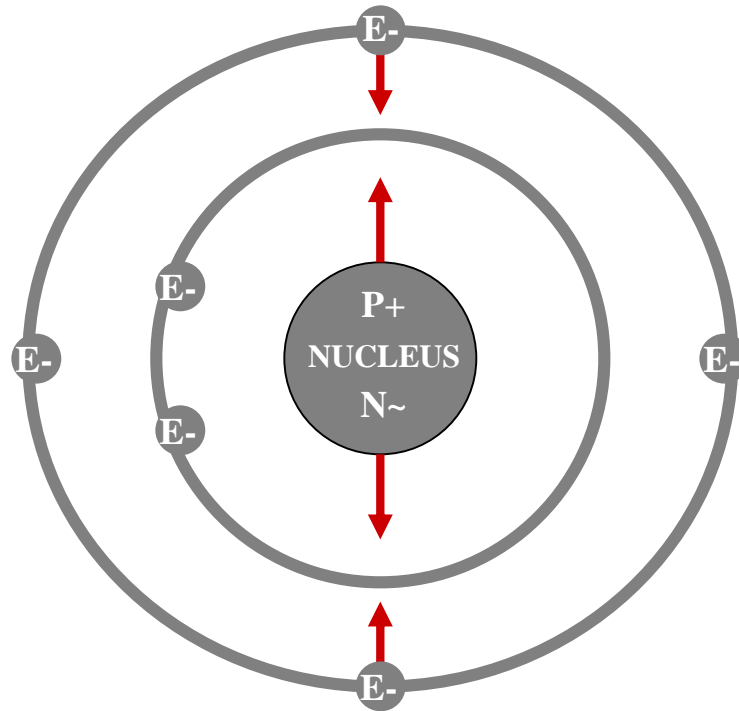
TYPICAL ATOM SUBATOMIC PARTICLES



TYPICAL ATOM SUBATOMIC PARTICLES

TYPICAL ATOM

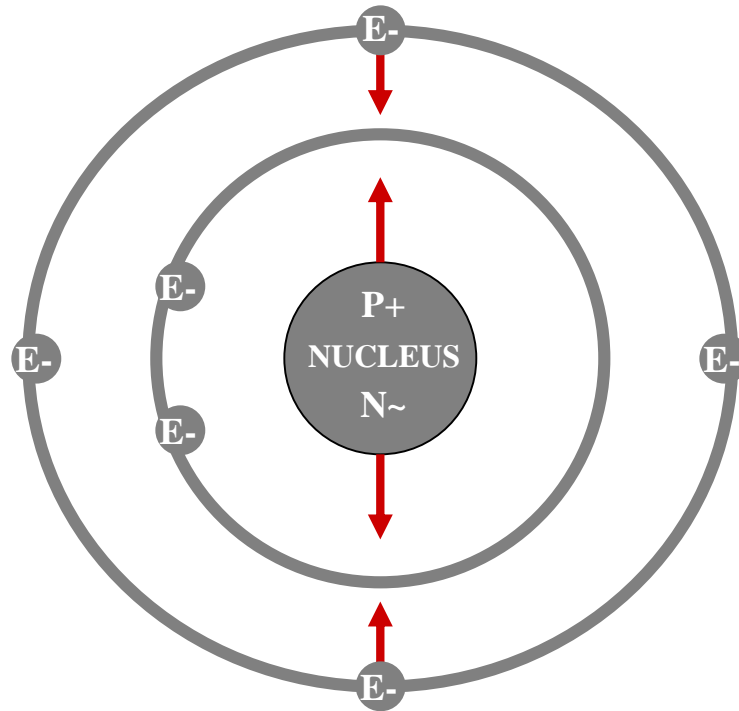
SUBATOMIC PARTICLES



P+ → ← E-

ATTRACT

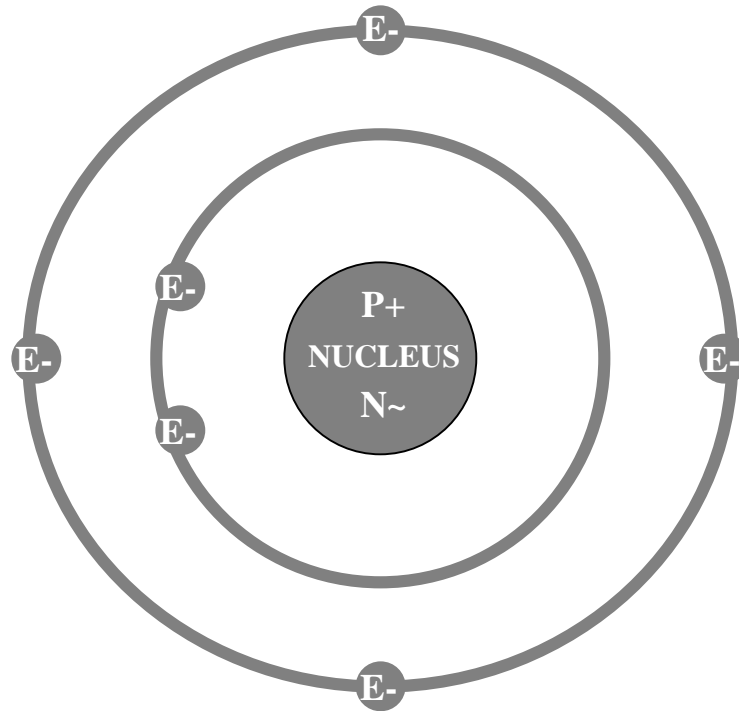
TYPICAL ATOM SUBATOMIC PARTICLES



P+ → ← E-

HOLDS ATOM INTACT

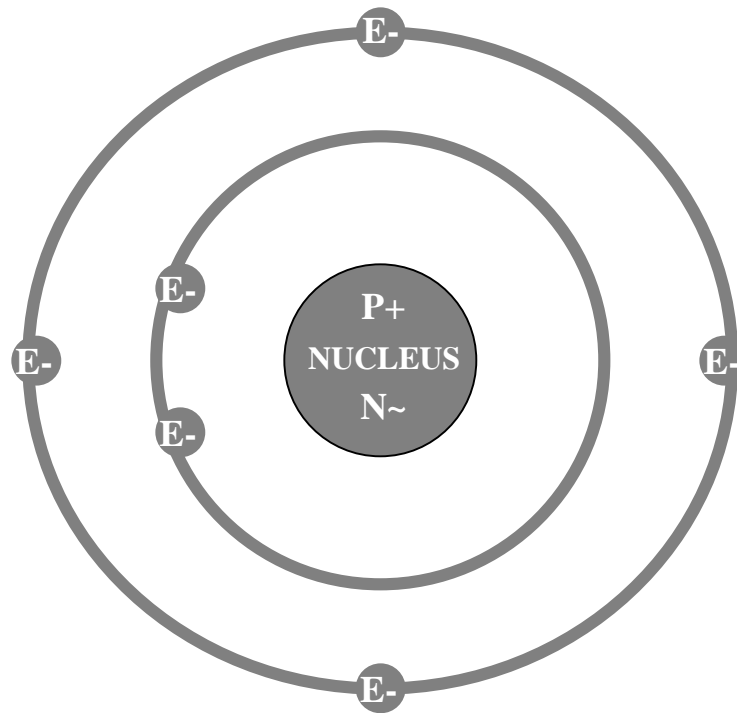
TYPICAL ATOM SUBATOMIC PARTICLES



TYPICAL ATOM ~ CHARGE

TYPICAL ATOM

SUBATOMIC PARTICLES

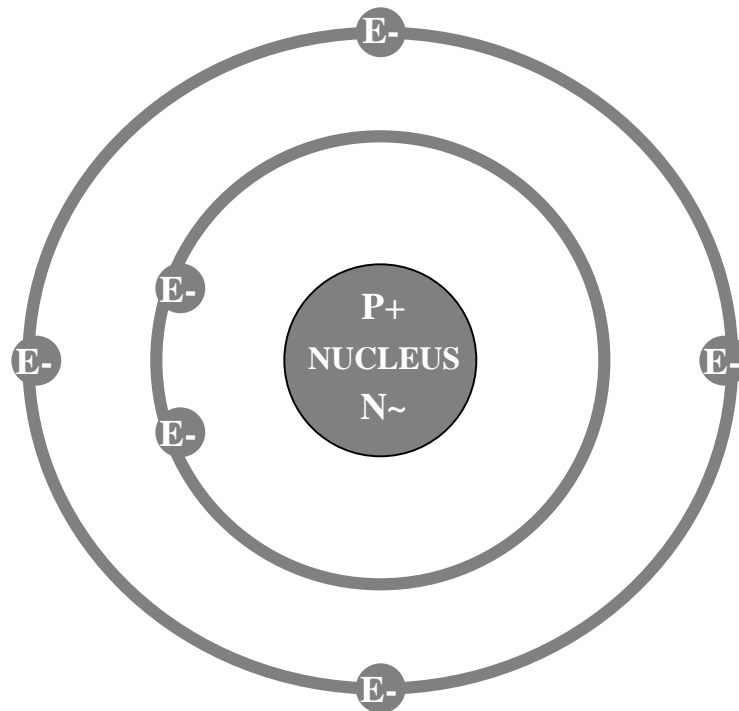


TYPICAL ATOM ~ CHARGE

NO. P+ NO. E-



TYPICAL ATOM SUBATOMIC PARTICLES



TYPICAL ATOM ~ CHARGE

NO. P+ = NO. E-

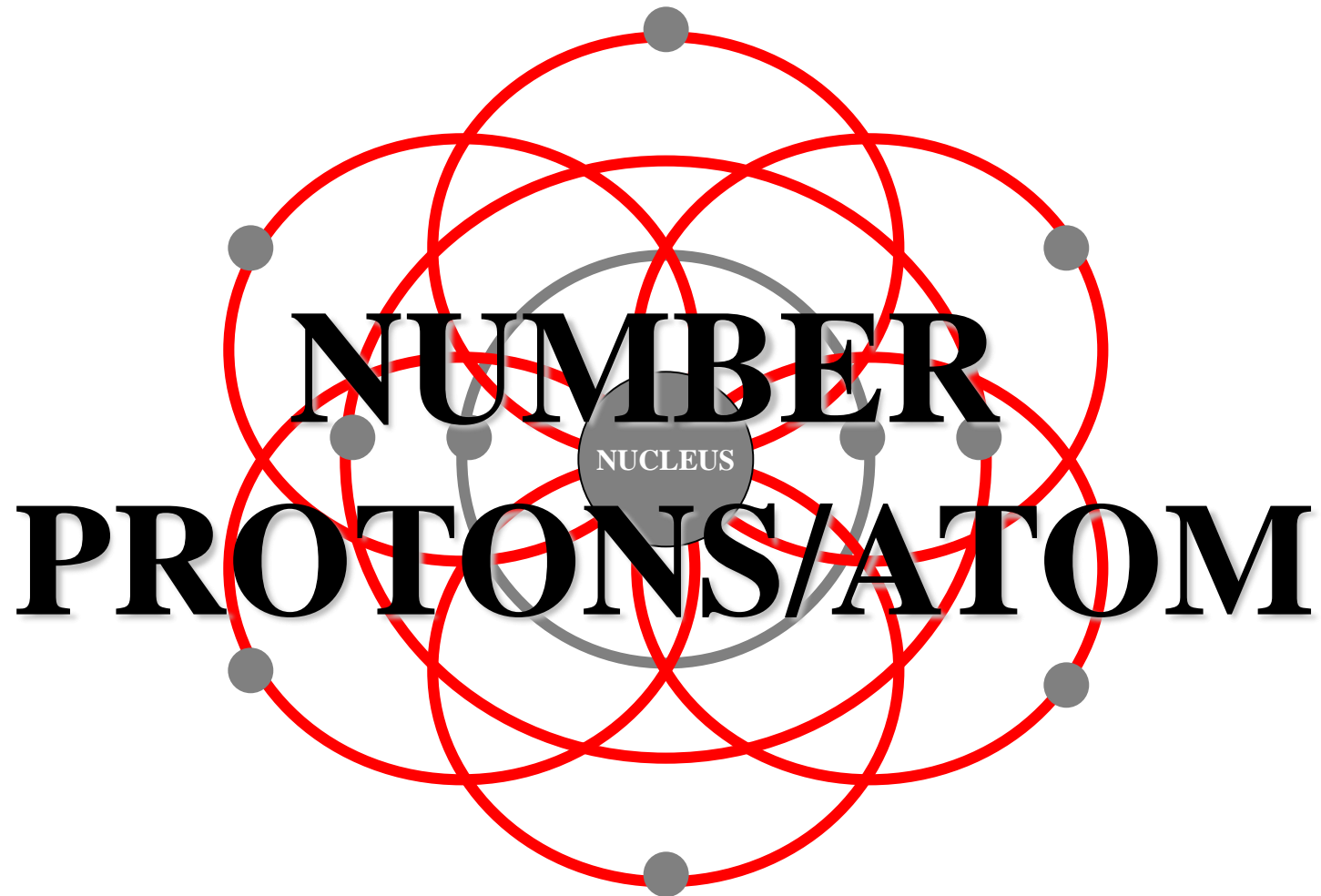


ATOMIC NO.
VS
ATOMIC WT.

ATOMIC NO.

ATOMIC NO.

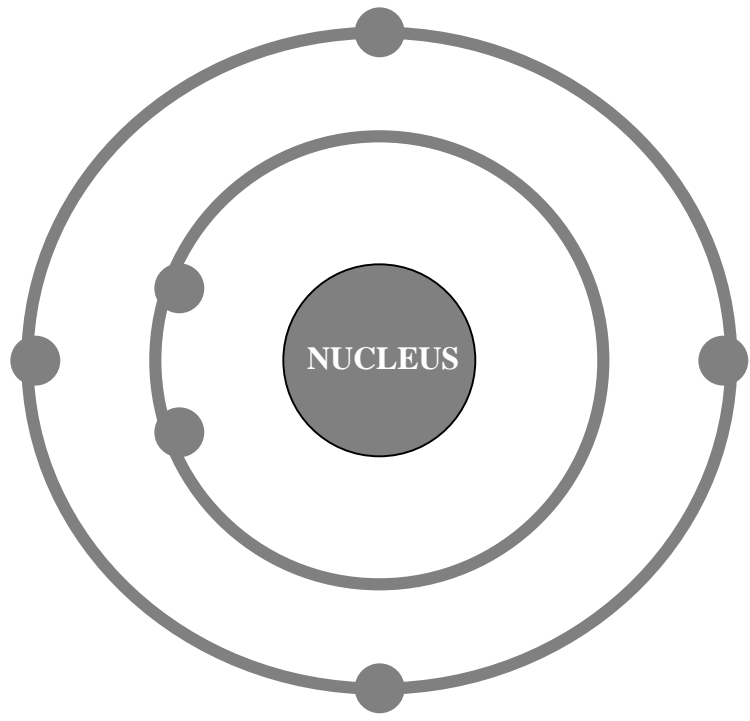
ATOMIC NUMBER



ATOMIC NUMBER

ATOMIC NO. EXAMPLE

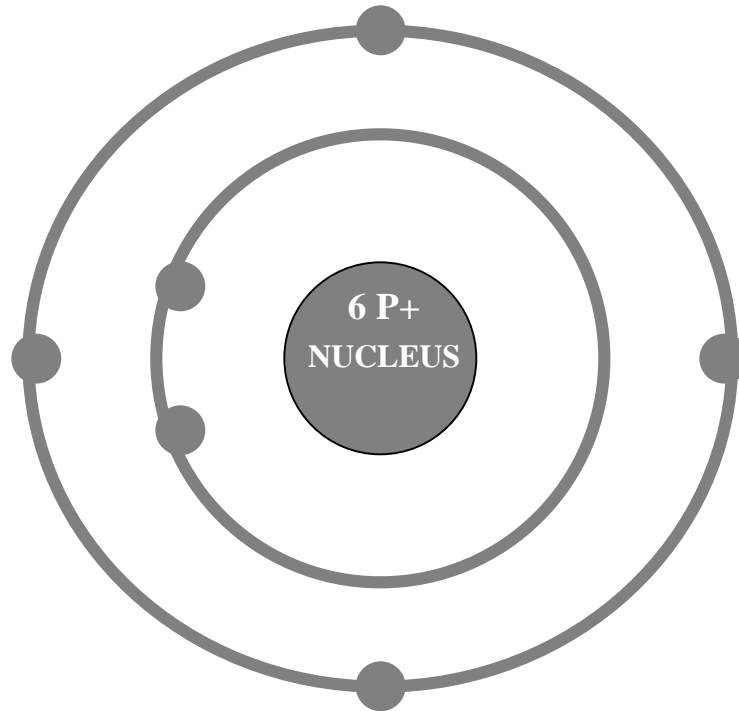
ATOMIC NUMBER



CARBON ATOM

● = e-

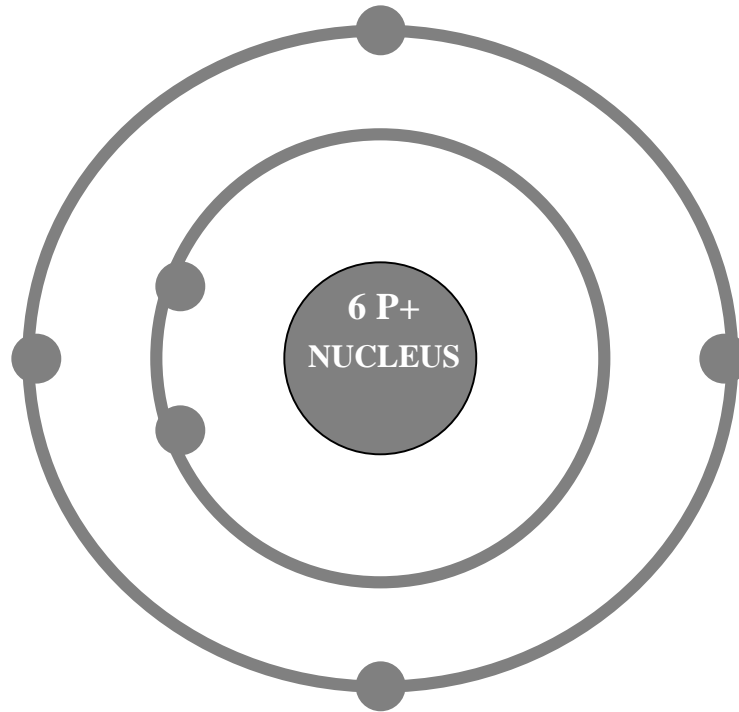
ATOMIC NUMBER



CARBON ATOM

● = e-

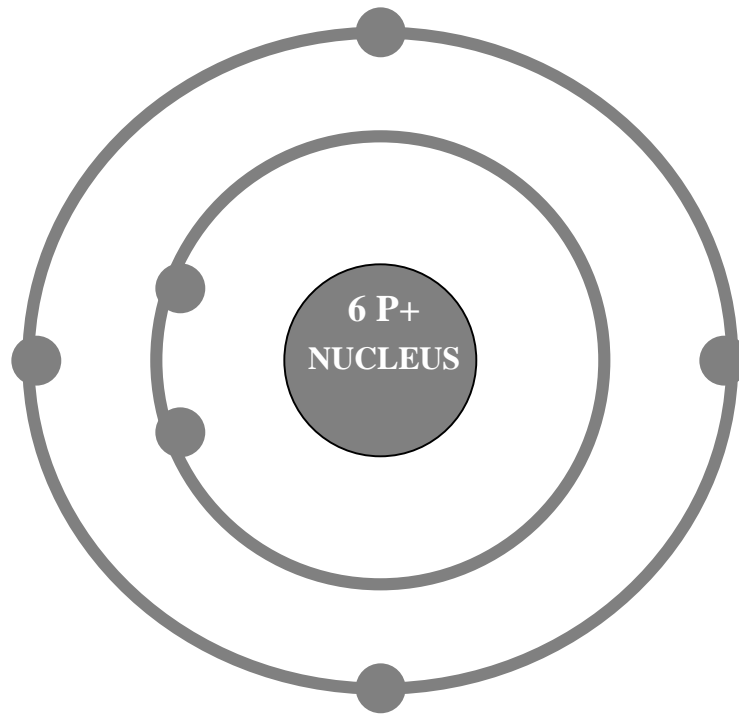
ATOMIC NUMBER



CARBON ATOMIC NO. = 6

● = e-

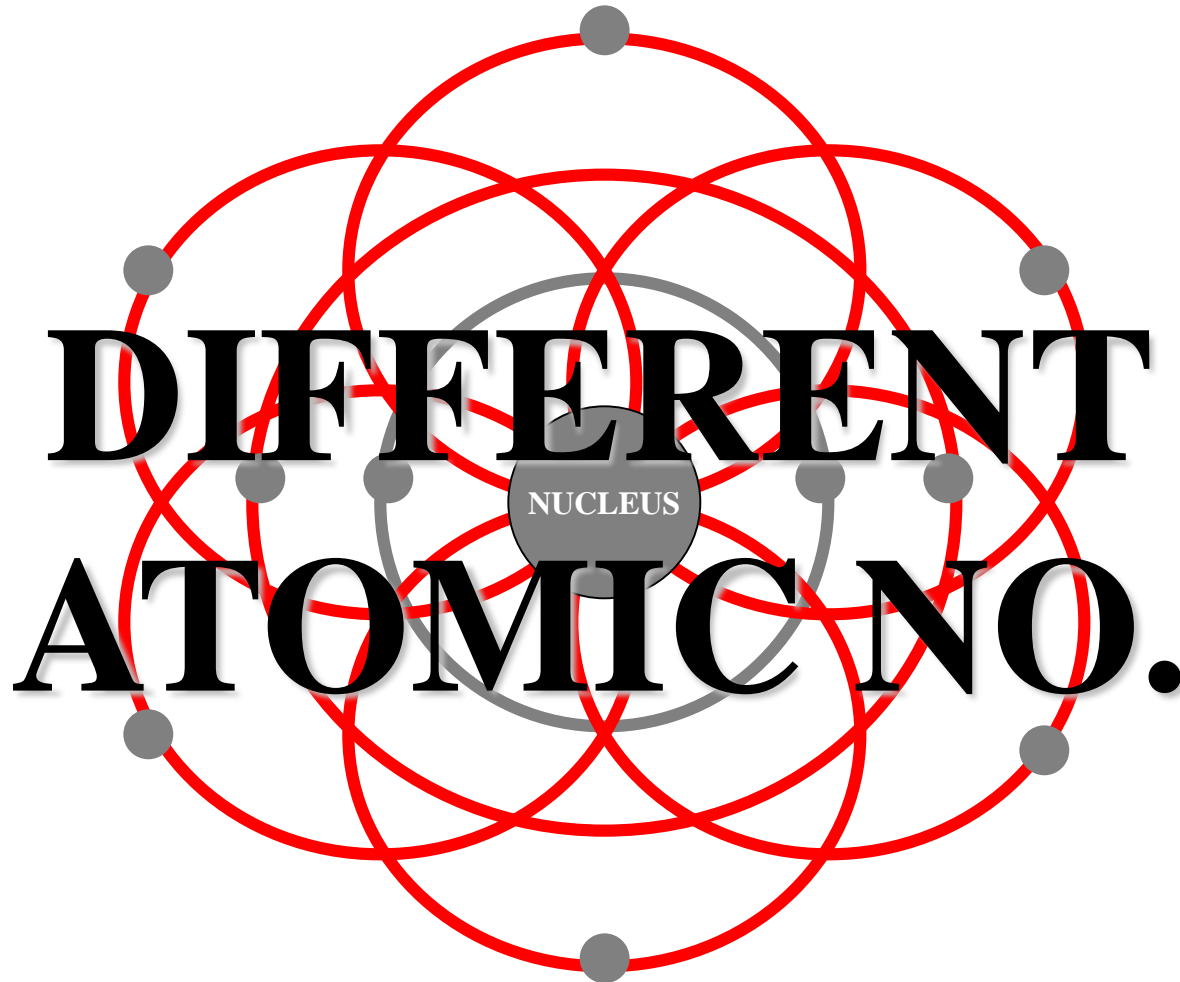
ATOMIC NUMBER



STABLE W/IN ELEMENT

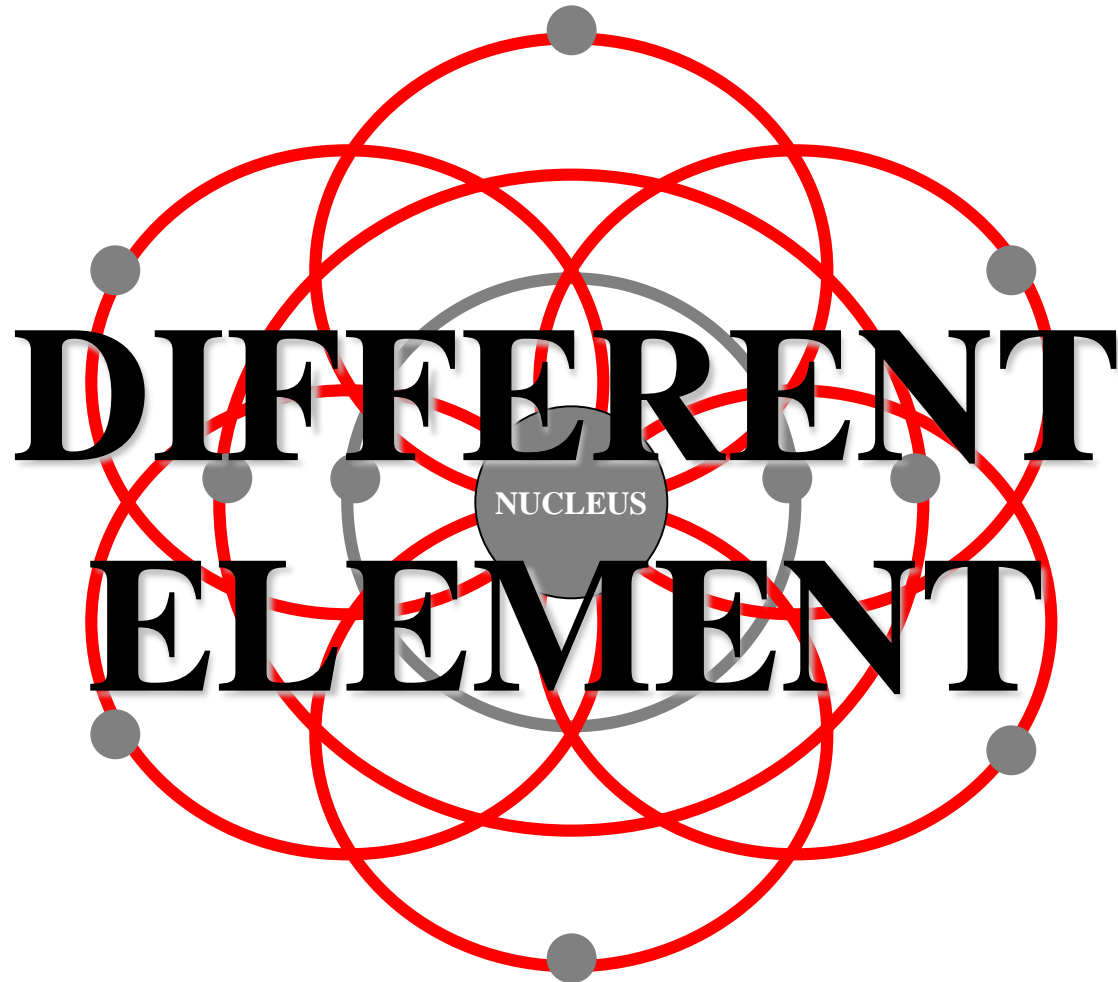
● = e-

ATOMIC NUMBER



ATOMIC NUMBER

ATOMIC NUMBER

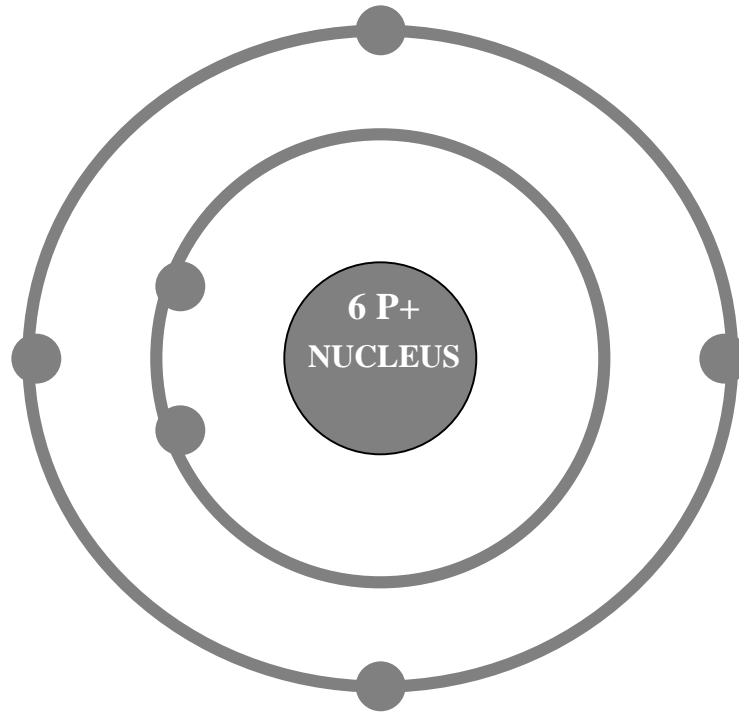


ATOMIC NUMBER



ATOMIC NO. APPLIED

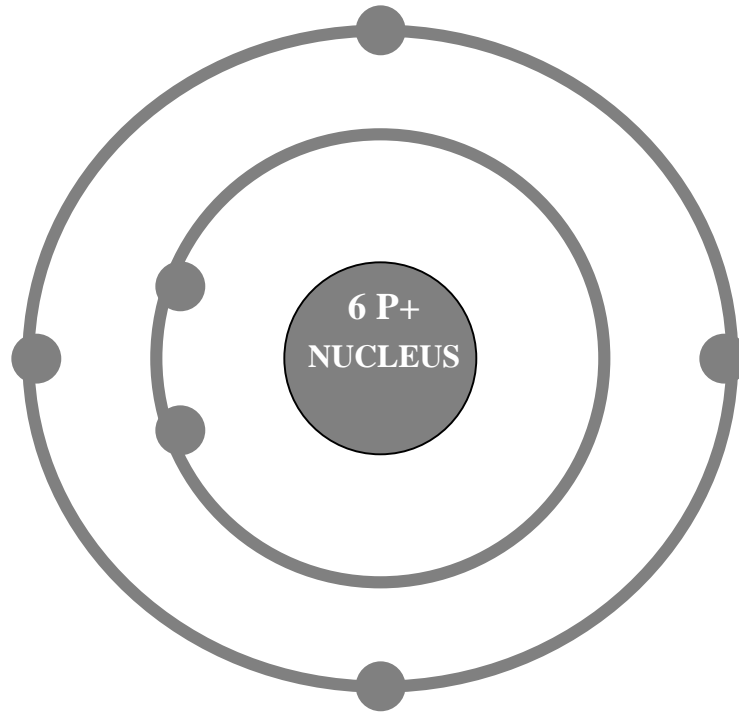
ATOMIC NUMBER



ATOMIC NO. = 6

● = e-

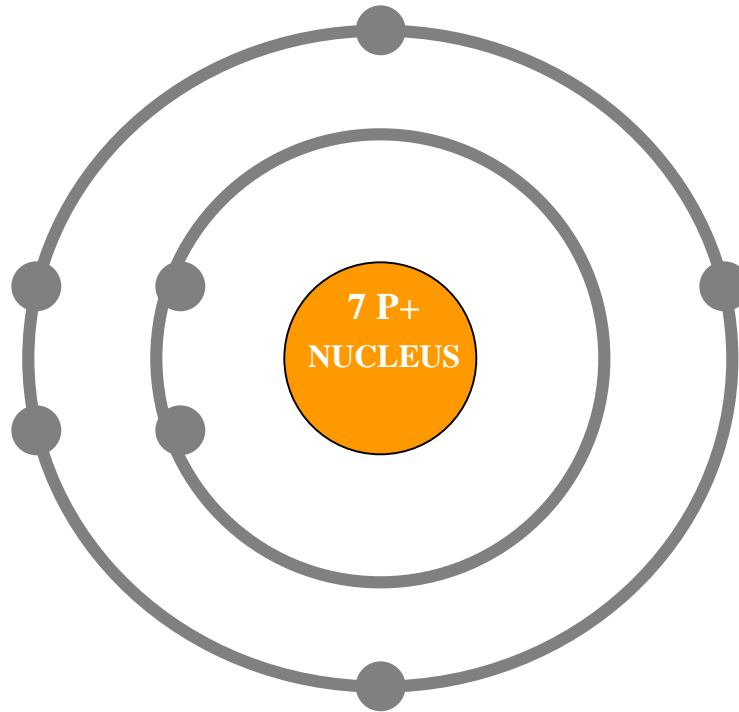
ATOMIC NUMBER



CARBON ATOM

● = e-

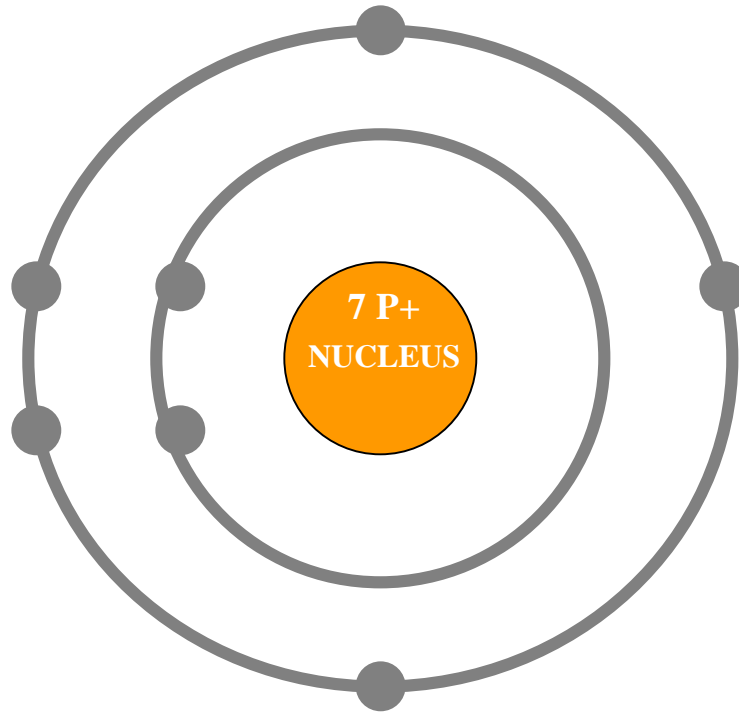
ATOMIC NUMBER



ATOMIC NO. = 7

● = E-

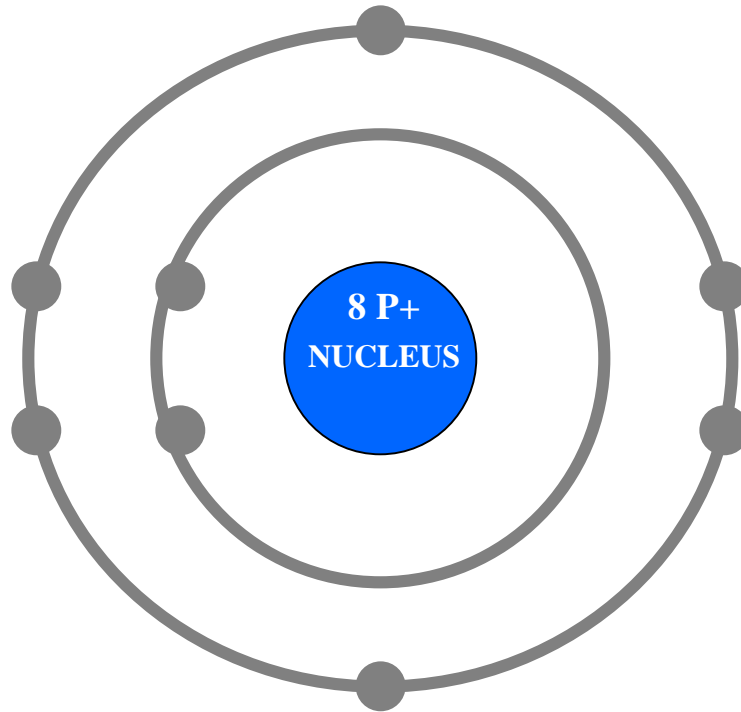
ATOMIC NUMBER



NITROGEN ATOM

● = e-

ATOMIC NUMBER



ATOMIC NO. = 8

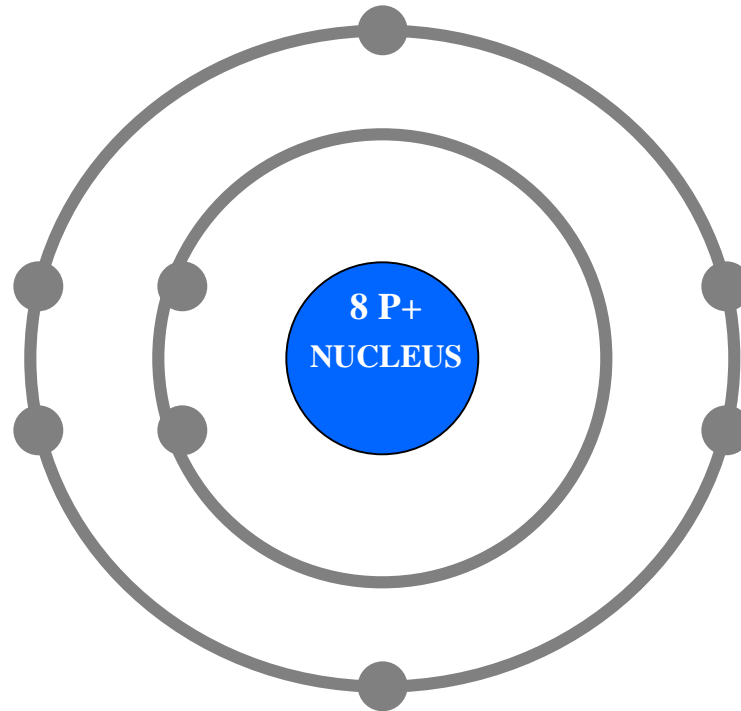
● = e-

ATOMIC NUMBER



D

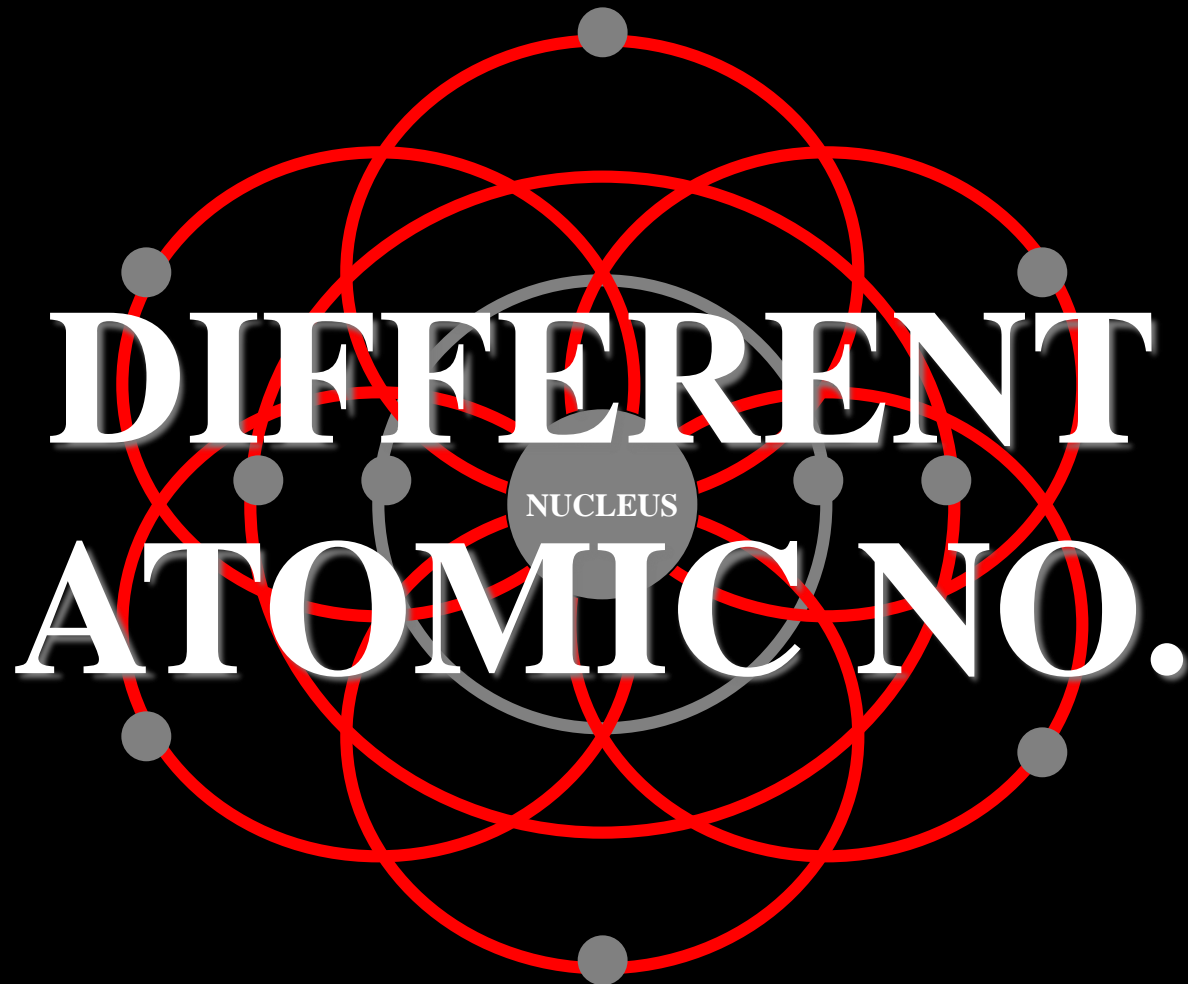
S



OXYGEN ATOM

● = E⁻

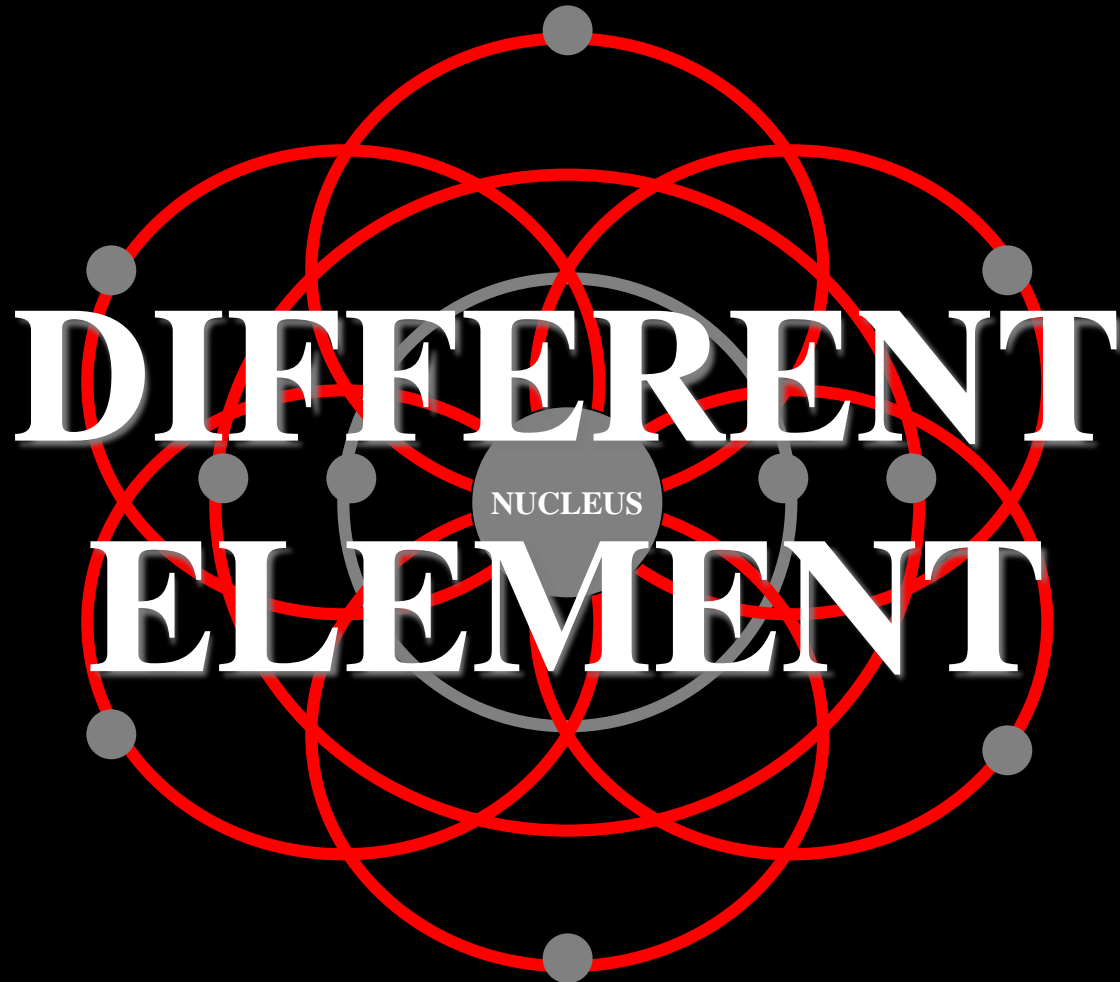
ATOMIC NUMBER



ATOMIC NUMBER



ATOMIC NUMBER



ATOMIC NUMBER

PERIODIC TABLE

1 H Hydrogen 1.00794																	2 He Helium 4.003				
3 Li Lithium 6.941	4 Be Beryllium 9.012182															5 B Boron 10.811	6 C Carbon 12.0107	7 N Nitrogen 14.00674	8 O Oxygen 15.9994	9 F Fluorine 18.9984032	10 Ne Neon 20.1797
11 Na Sodium 22.989770	12 Mg Magnesium 24.3050															13 Al Aluminum 26.981538	14 Si Silicon 28.0855	15 P Phosphorus 30.973761	16 S Sulfur 32.066	17 Cl Chlorine 35.4527	18 Ar Argon 39.948
19 K Potassium 39.0983	20 Ca Calcium 40.078	21 Sc Scandium 44.955910	22 Ti Titanium 47.867	23 V Vanadium 50.9415	24 Cr Chromium 51.9961	25 Mn Manganese 54.938049	26 Fe Iron 55.845	27 Co Cobalt 58.933200	28 Ni Nickel 58.6934	29 Cu Copper 63.546	30 Zn Zinc 65.39	31 Ga Gallium 69.723	32 Ge Germanium 72.61	33 As Arsenic 74.92160	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80				
37 Rb Rubidium 85.4678	38 Sr Strontium 87.62	39 Y Yttrium 88.90585	40 Zr Zirconium 91.224	41 Nb Niobium 92.90638	42 Mo Molybdenum 95.94	43 Tc Technetium (98)	44 Ru Ruthenium 101.07	45 Rh Rhodium 102.90550	46 Pd Palladium 106.42	47 Ag Silver 107.8682	48 Cd Cadmium 112.411	49 In Indium 114.818	50 Sn Tin 118.710	51 Sb Antimony 121.760	52 Te Tellurium 127.60	53 I Iodine 126.90447	54 Xe Xenon 131.29				
55 Cs Cesium 132.90545	56 Ba Barium 137.327	57 La Lanthanum 138.9055	72 Hf Hafnium 178.49	73 Ta Tantalum 180.9479	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.217	78 Pt Platinum 195.078	79 Au Gold 196.96655	80 Hg Mercury 200.59	81 Tl Thallium 204.3833	82 Pb Lead 207.2	83 Bi Bismuth 208.98038	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222)				
87 Fr Francium (223)	88 Ra Radium (226)	89 Ac Actinium (227)	104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (263)	107 Bh Bohrium (262)	108 Hs Hassium (265)	109 Mt Meitnerium (266)	110 (269)	111 (272)	112 (277)	113	114								

58 Ce Cerium 140.116	59 Pr Praseodymium 140.90765	60 Nd Neodymium 144.24	61 Pm Promethium (145)	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.92534	66 Dy Dysprosium 162.50	67 Ho Holmium 164.93032	68 Er Erbium 167.26	69 Tm Thulium 168.93421	70 Yb Ytterbium 173.04	71 Lu Lutetium 174.967
90 Th Thorium 232.0381	91 Pa Protactinium 231.03588	92 U Uranium 238.0289	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)



PERIODIC TABLE

ARRANGED BY ATOMIC NO.

1 H Hydrogen 1.00794																	2 He Helium 4.003						
3 Li Lithium 6.941	4 Be Beryllium 9.012182																	5 B Boron 10.811	6 C Carbon 12.0107	7 N Nitrogen 14.00674	8 O Oxygen 15.9994	9 F Fluorine 18.9984032	10 Ne Neon 20.1797
11 Na Sodium 22.989770	12 Mg Magnesium 24.3050																	13 Al Aluminum 26.981538	14 Si Silicon 28.0855	15 P Phosphorus 30.973761	16 S Sulfur 32.066	17 Cl Chlorine 35.4527	18 Ar Argon 39.948
19 K Potassium 39.0983	20 Ca Calcium 40.078	21 Sc Scandium 44.955912	22 Ti Titanium 47.88	23 V Vanadium 50.9415	24 Cr Chromium 51.9961	25 Mn Manganese 54.938	26 Fe Iron 55.845	27 Co Cobalt 58.933200	28 Ni Nickel 58.6934	29 Cu Copper 63.546	30 Zn Zinc 65.39	31 Ga Gallium 69.723	32 Ge Germanium 72.64	33 As Arsenic 74.9216	34 Se Selenium 78.96	35 Br Bromine 79.904	36 Kr Krypton 83.80						
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55 Cs Cesium 132.90545	56 Ba Barium 137.327	57 La Lanthanum 138.9055	72 Hf Hafnium 178.49	73 Ta Tantalum 180.9479	74 W Tungsten 183.84	75 Re Rhenium 186.207	76 Os Osmium 190.23	77 Ir Iridium 192.222	78 Pt Platinum 195.084	79 Au Gold 196.96655	80 Hg Mercury 200.59	81 Tl Thallium 204.3833	82 Pb Lead 207.2	83 Bi Bismuth 208.98038	84 Po Polonium (209)	85 At Astatine (210)	86 Rn Radon (222)						
87 Fr Francium (223)	88 Ra Radium (226)	89 Ac Actinium (227)	104 Rf Rutherfordium (261)	105 Db Dubnium (262)	106 Sg Seaborgium (263)	107 Bh Bohrium (262)	108 Hs Hassium (265)	109 Mt Meitnerium (266)	110 Ds Darmstadtium (269)	111 Rg Roentgenium (272)	112 Cn Copernicium (277)	113 Nh Nihonium (284)	114 Fl Flerovium (289)	115 Mc Moscovium (288)	116 Lv Livermorium (293)	117 Ts Tennessine (294)	118 Og Oganesson (294)						

58 Ce Cerium 140.116	59 Pr Praseodymium 140.90765	60 Nd Neodymium 144.24	61 Pm Promethium (145)	62 Sm Samarium 150.36	63 Eu Europium 151.964	64 Gd Gadolinium 157.25	65 Tb Terbium 158.92534	66 Dy Dysprosium 162.50	67 Ho Holmium 164.93032	68 Er Erbium 167.26	69 Tm Thulium 168.93421	70 Yb Ytterbium 173.04	71 Lu Lutetium 174.967
90 Th Thorium 232.0381	91 Pa Protactinium 231.03588	92 U Uranium 238.0289	93 Np Neptunium (237)	94 Pu Plutonium (244)	95 Am Americium (243)	96 Cm Curium (247)	97 Bk Berkelium (247)	98 Cf Californium (251)	99 Es Einsteinium (252)	100 Fm Fermium (257)	101 Md Mendelevium (258)	102 No Nobelium (259)	103 Lr Lawrencium (262)



ATOMIC NUMBER



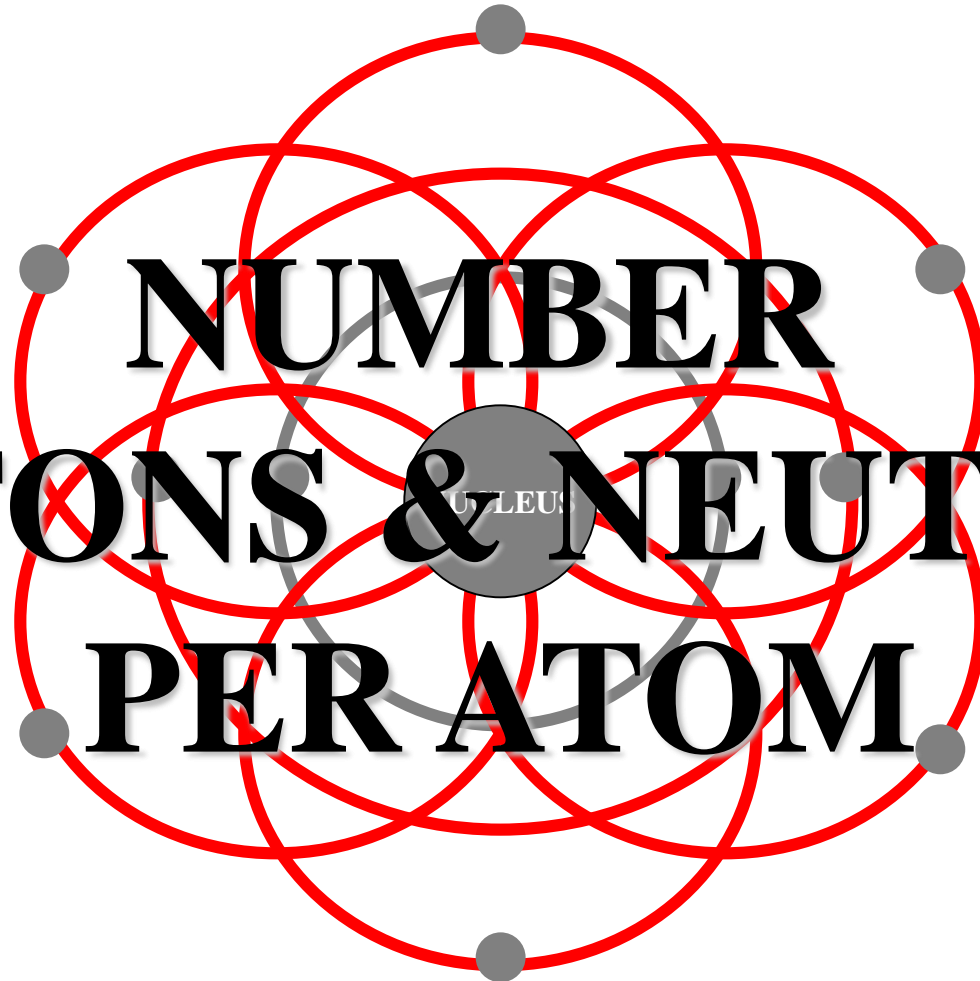
**ATOMIC NO.
DOES NOT VARY
/ATOM/ELEMENT**

ATOMIC NUMBER

ATOMIC WT.

ATOMIC WT.

ATOMIC WEIGHT



NUMBER

OF PROTONS & NEUTRONS

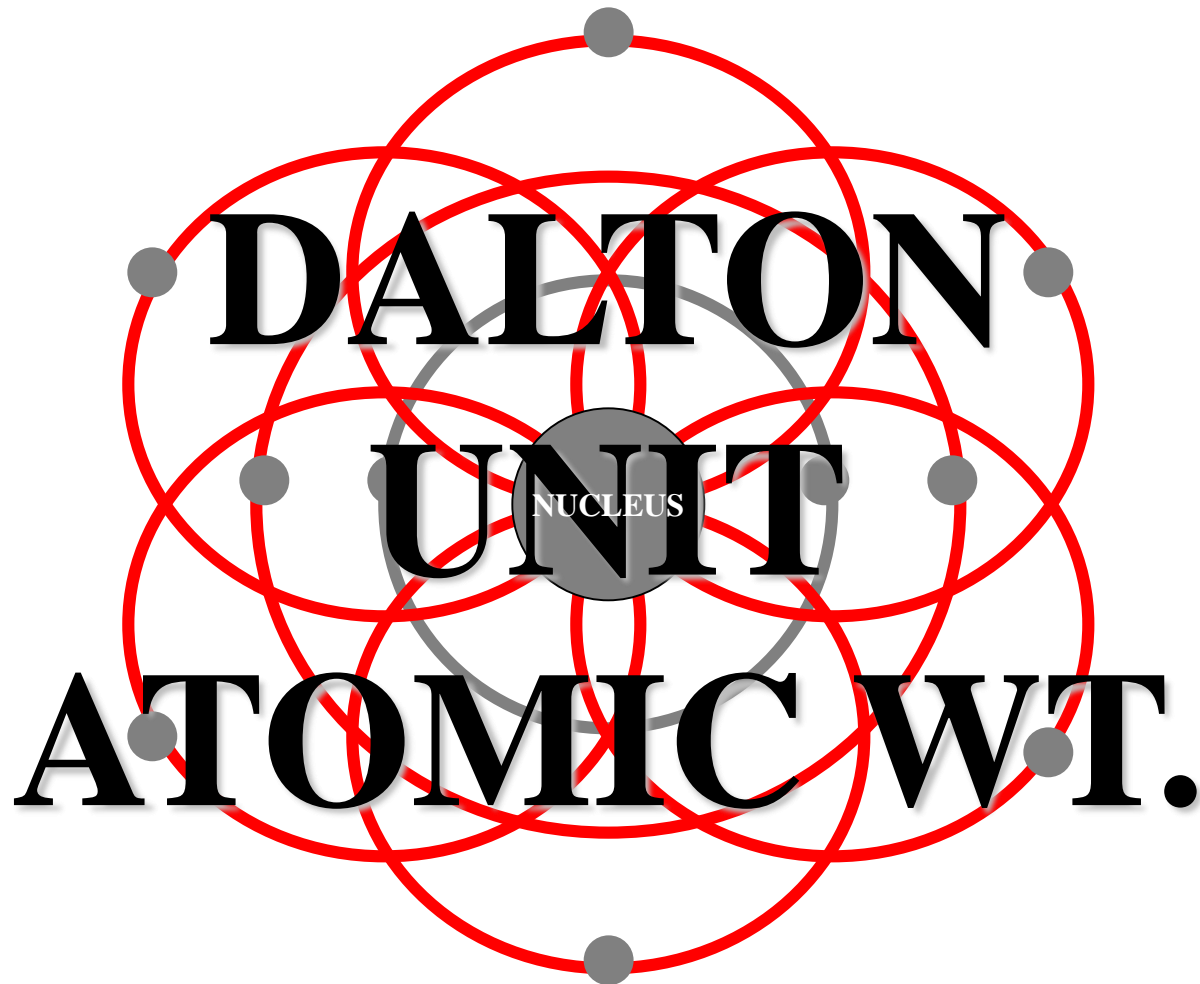
PER ATOM

ATOMIC WEIGHT

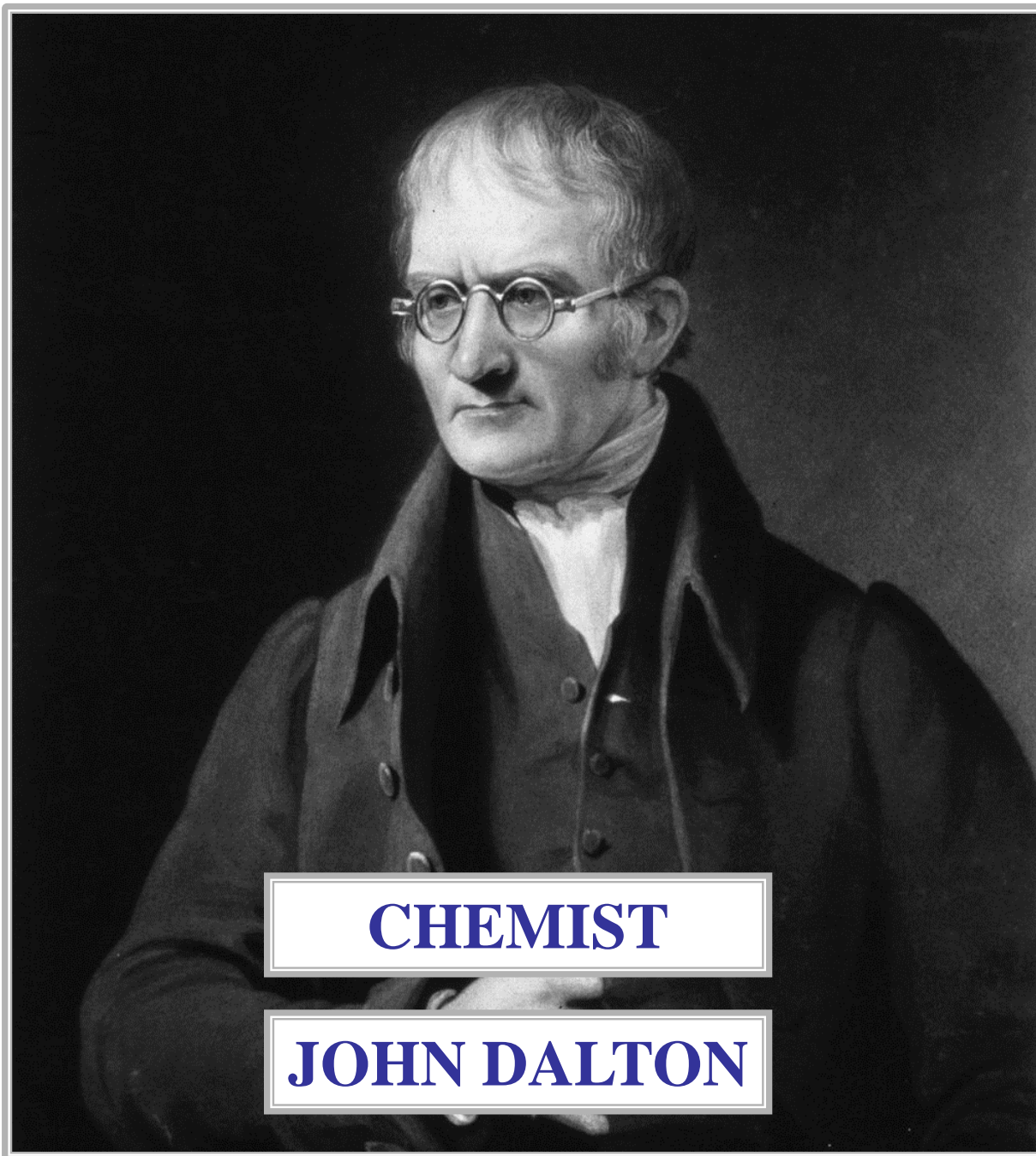
DALTON



ATOMIC WEIGHT



ATOMIC WEIGHT



CHEMIST

JOHN DALTON

DALTON
ATOMIC WEIGHT

PROTON

DALTON
ATOMIC WEIGHT

PROTON = 1 DALTON



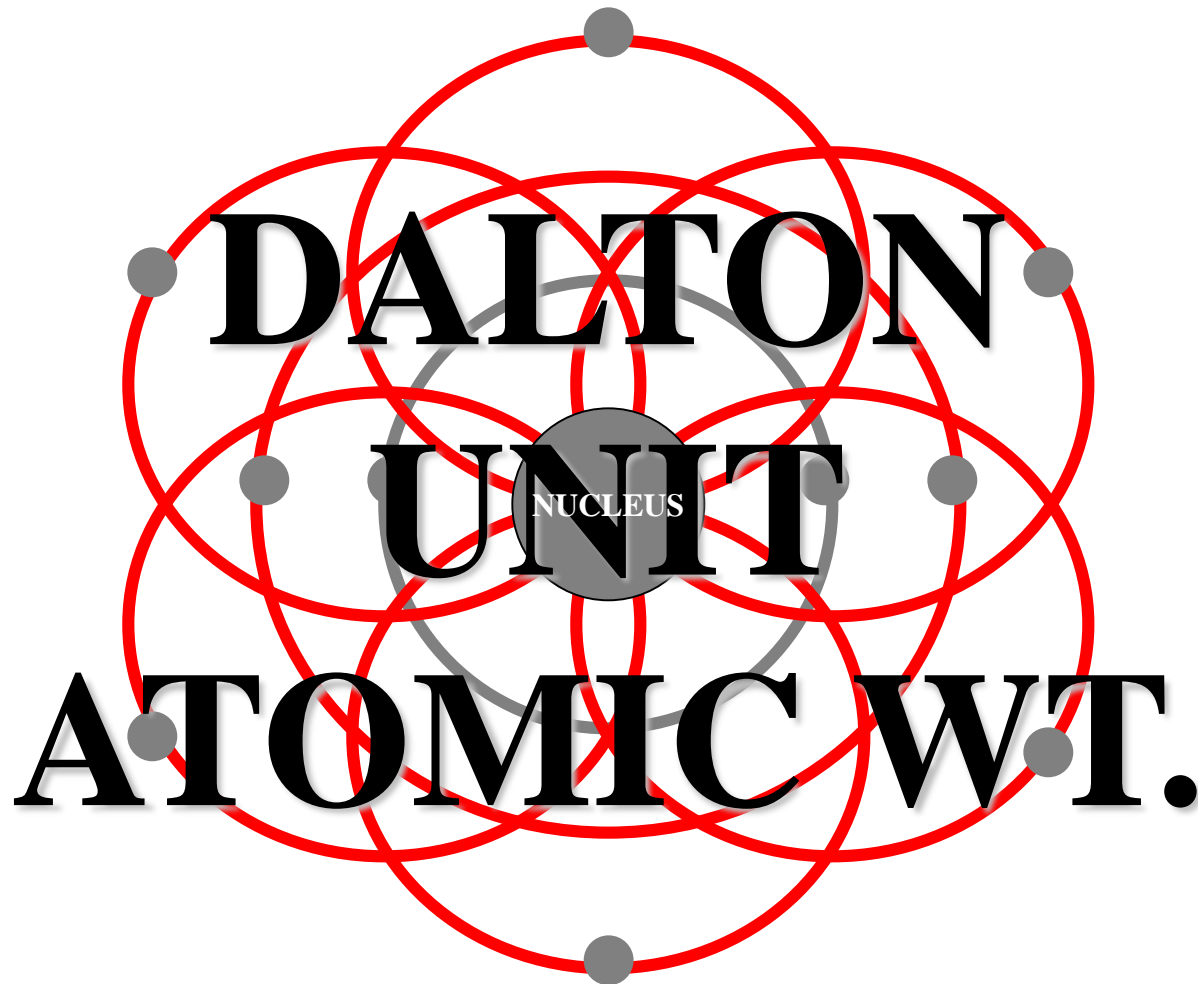
PROTON = 1 DALTON

DALTON
ATOMIC WEIGHT

NEUTRON

DALTON
ATOMIC WEIGHT

NEUTRON = 1 DALTON



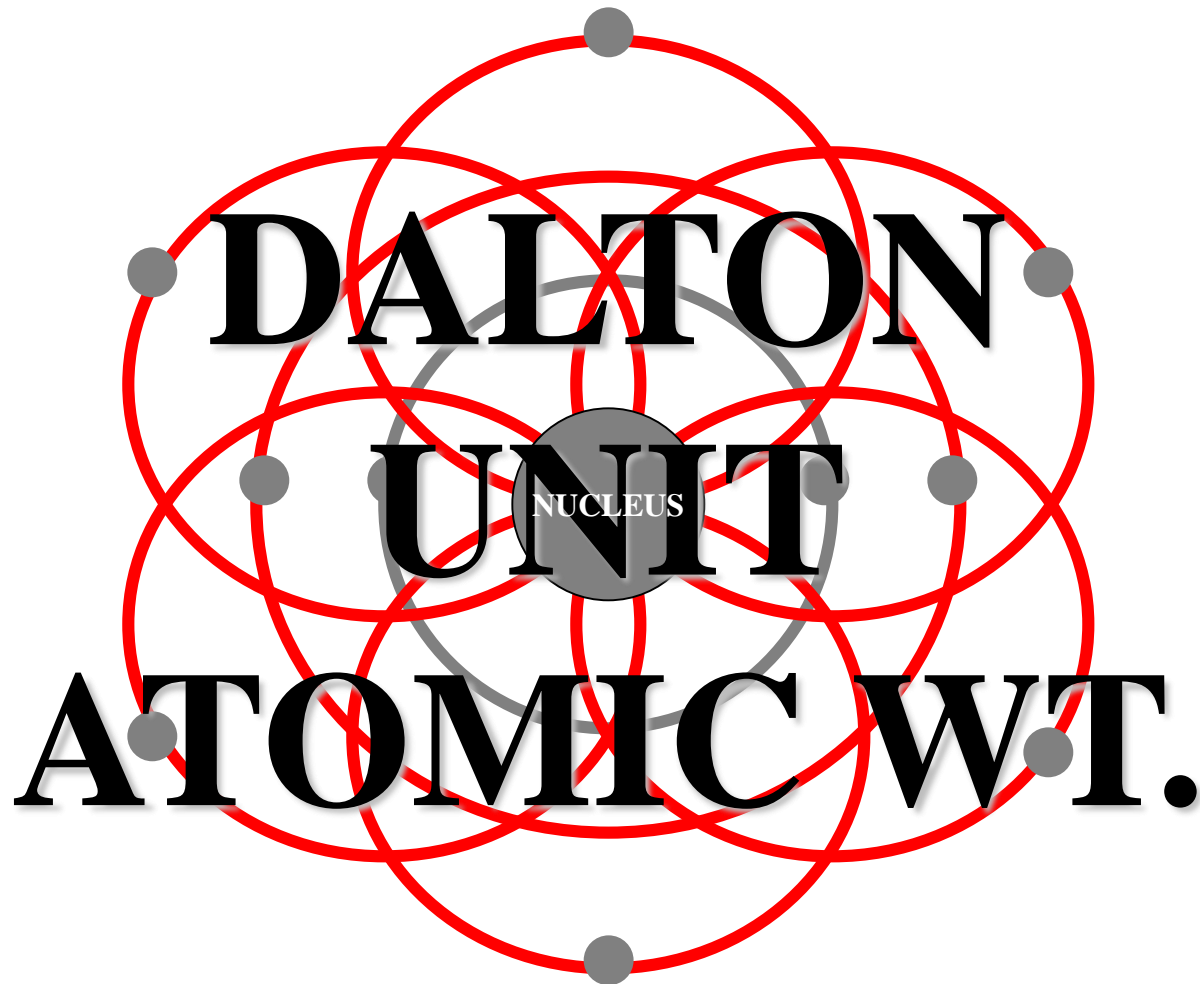
NEUTRON = 1 DALTON

**DALTON
ATOMIC WEIGHT**

ELECTRON

**DALTON
ATOMIC WEIGHT**

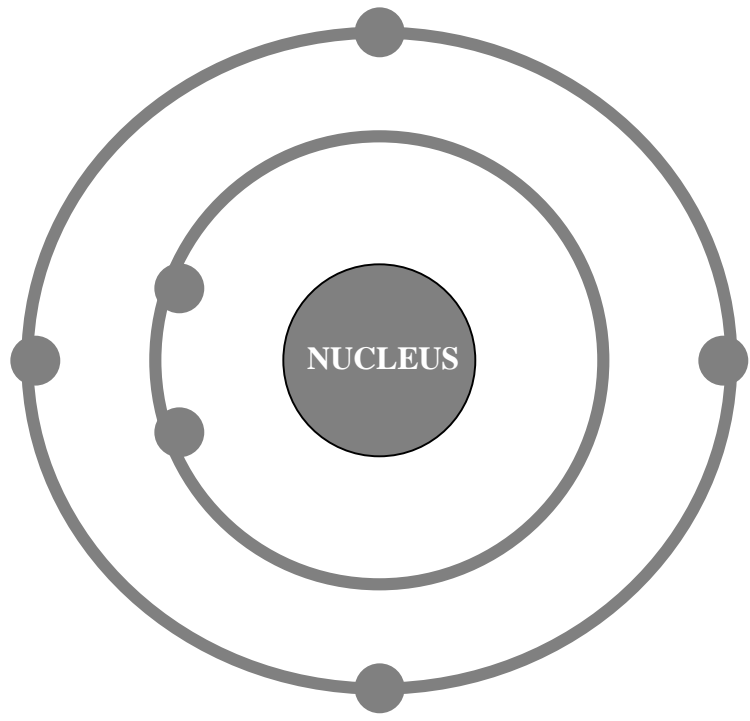
ELECTRON = 0 DALTON



ELECTRON = 0 DALTON

ATOMIC WT. EXAMPLE

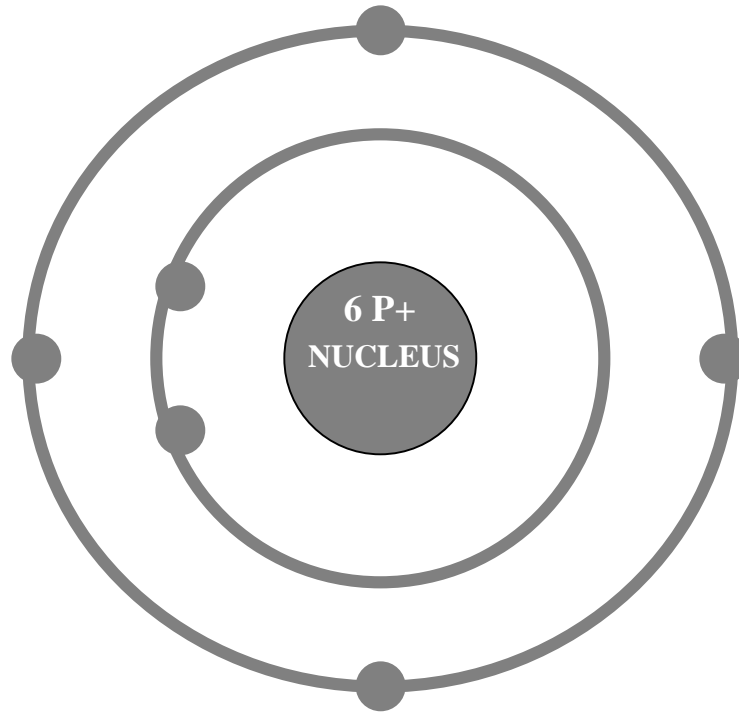
ATOMIC WEIGHT



TYPICAL CARBON ATOM

● = e-

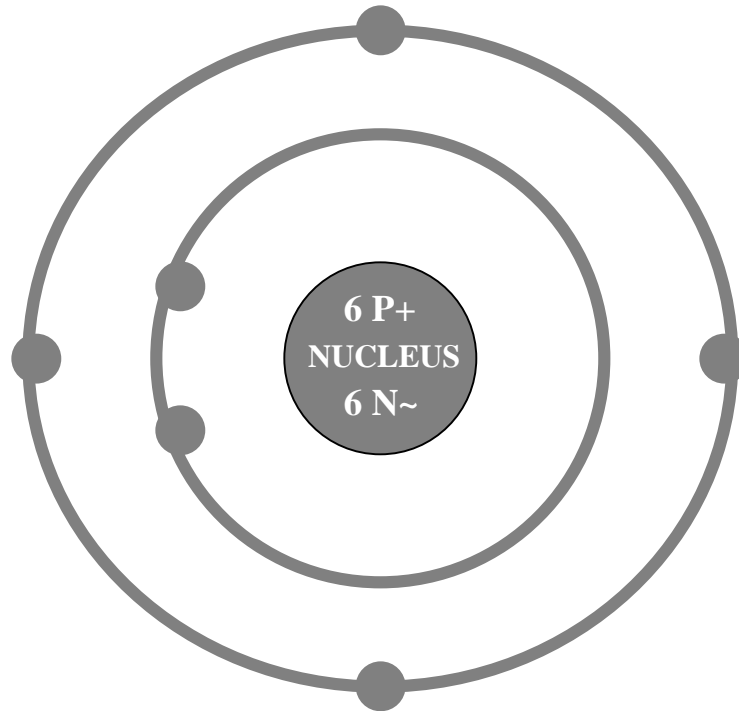
ATOMIC WEIGHT



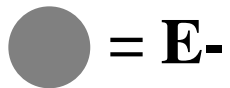
TYPICAL CARBON ATOM

● = e-

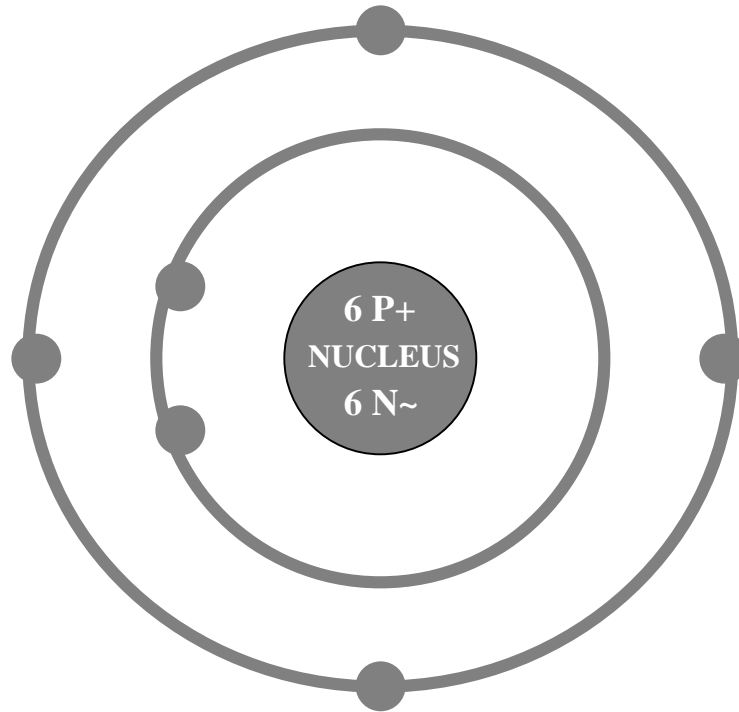
ATOMIC WEIGHT



TYPICAL CARBON ATOM



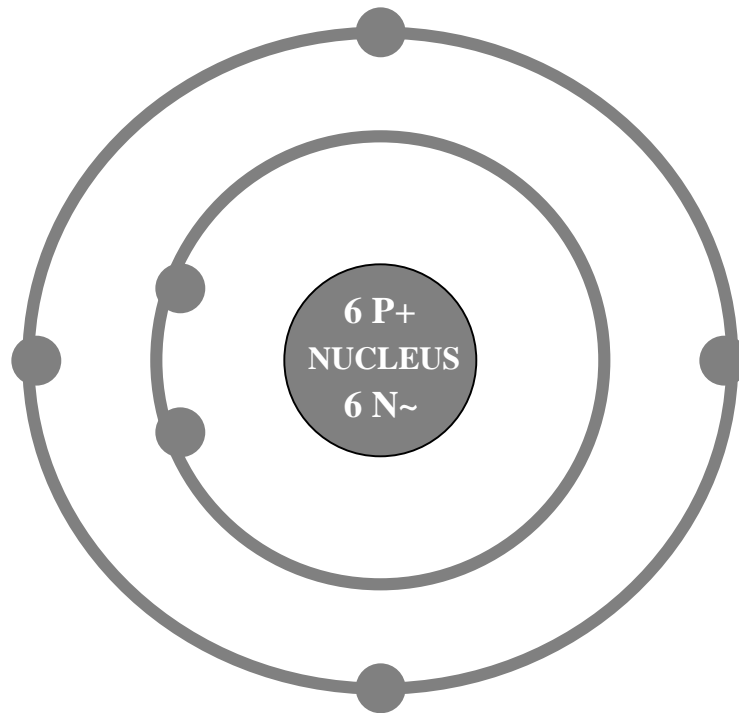
ATOMIC WEIGHT



CARBON AT. WT. = 12 DALTONS

● = e⁻

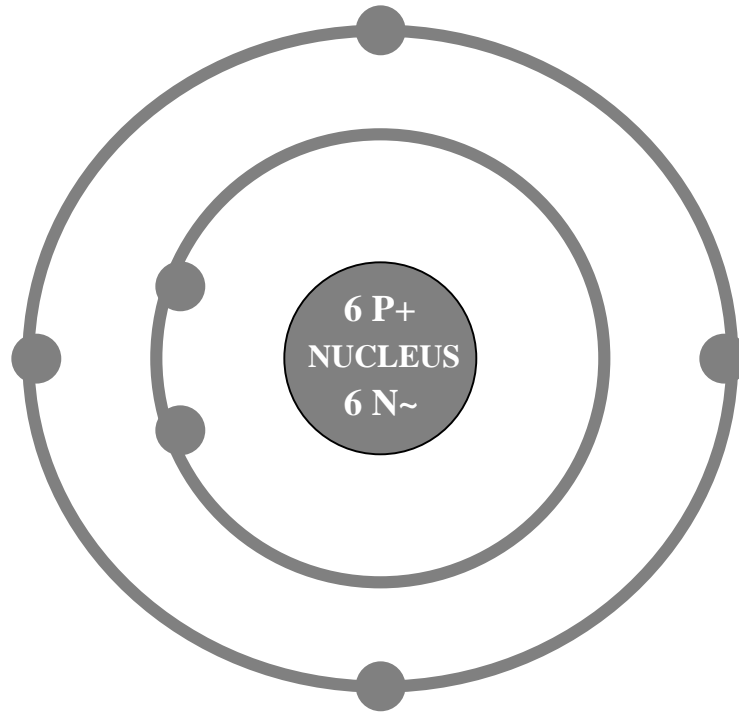
ATOMIC WEIGHT



PROTON NO./ATOM/ELEMENT

● = E-

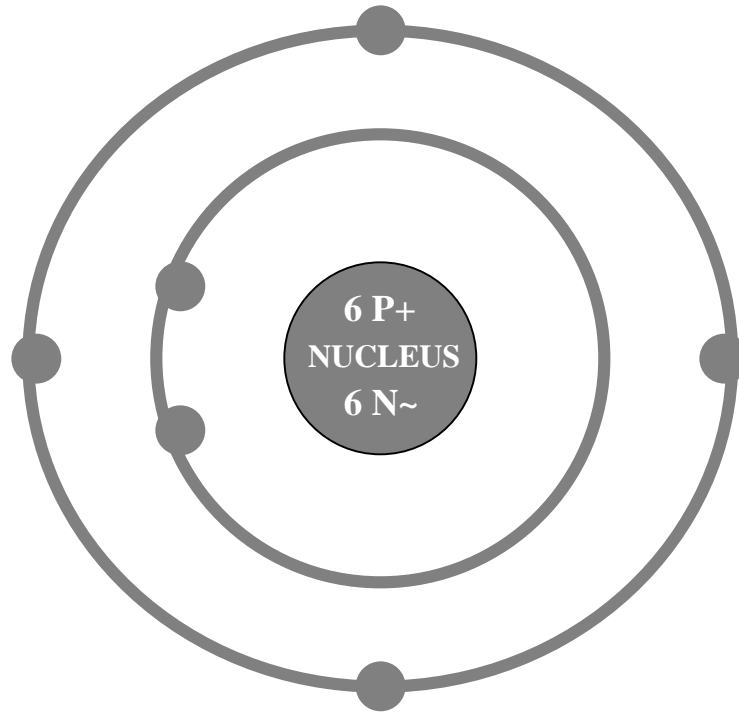
ATOMIC WEIGHT



PROTON NO. STABLE

● = e-

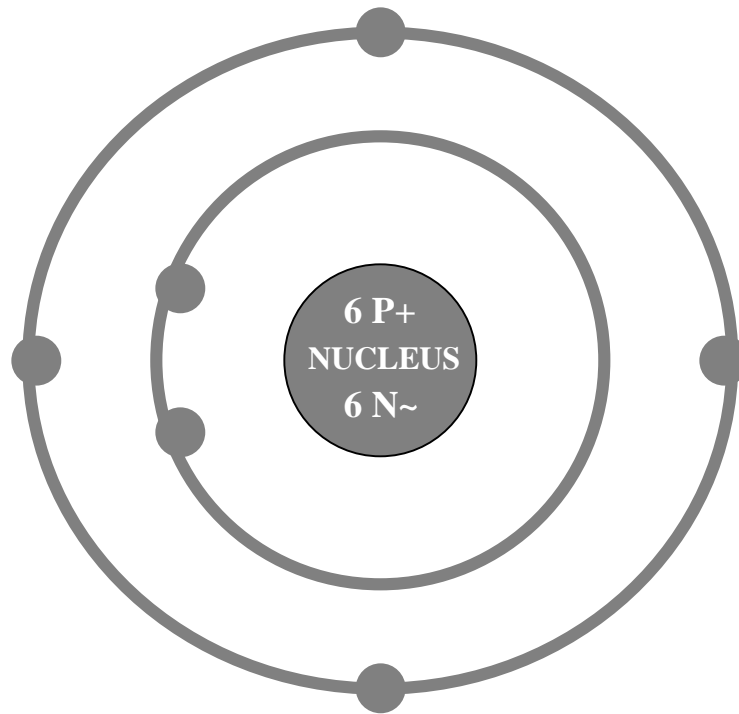
ATOMIC WEIGHT



NEUTRON NO./ATOM/ELEMENT

● = E-

ATOMIC WEIGHT

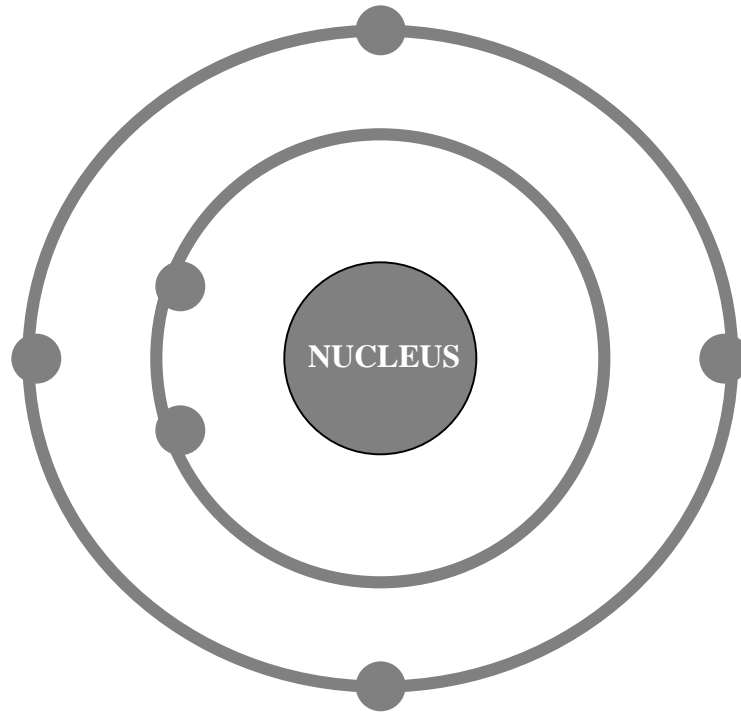


NEUTRON NO. MAY VARY

● = e-

**NEUTRON NO.
MAY VARY
EXAMPLE**

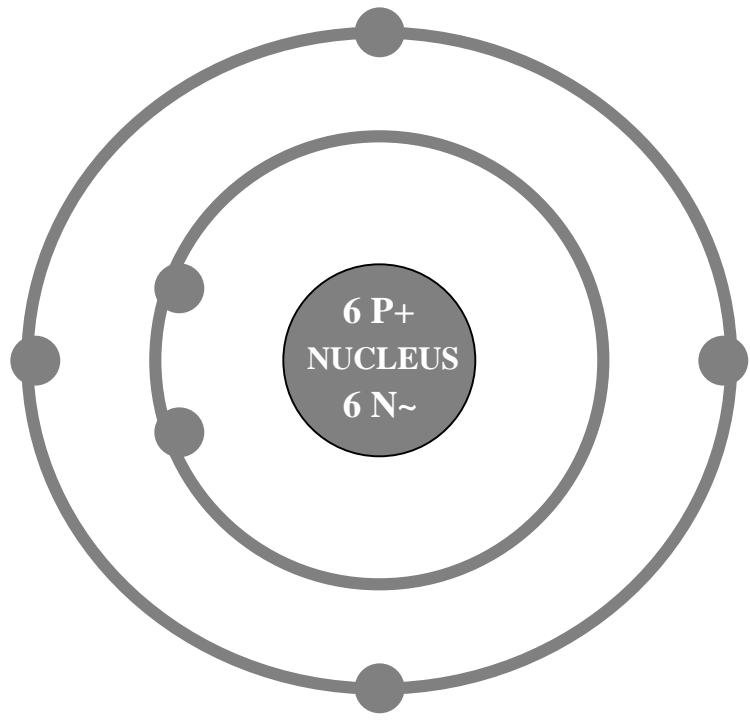
ATOMIC WEIGHT



TYPICAL CARBON ATOM

● = e-

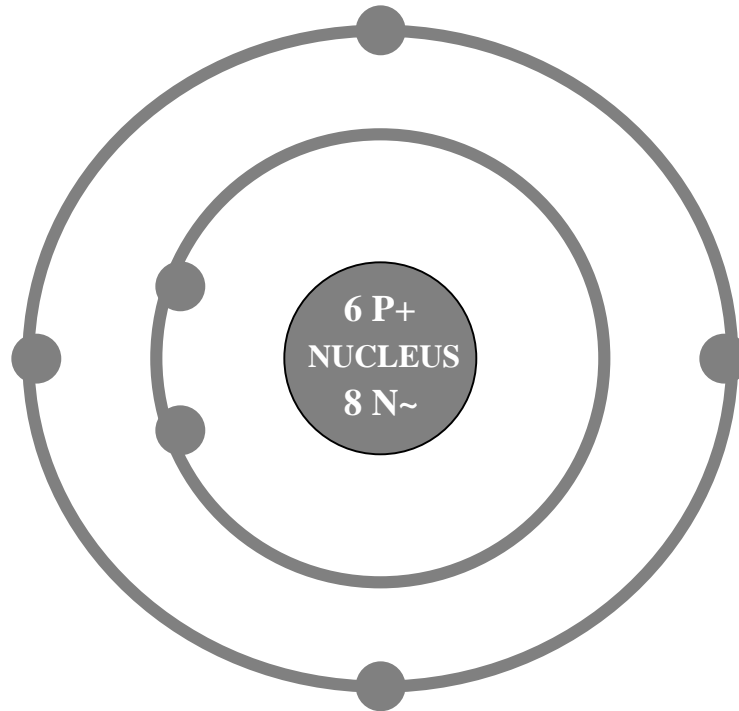
ATOMIC WEIGHT



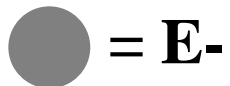
TYPICAL CARBON ATOM

● = e-

ATOMIC WEIGHT

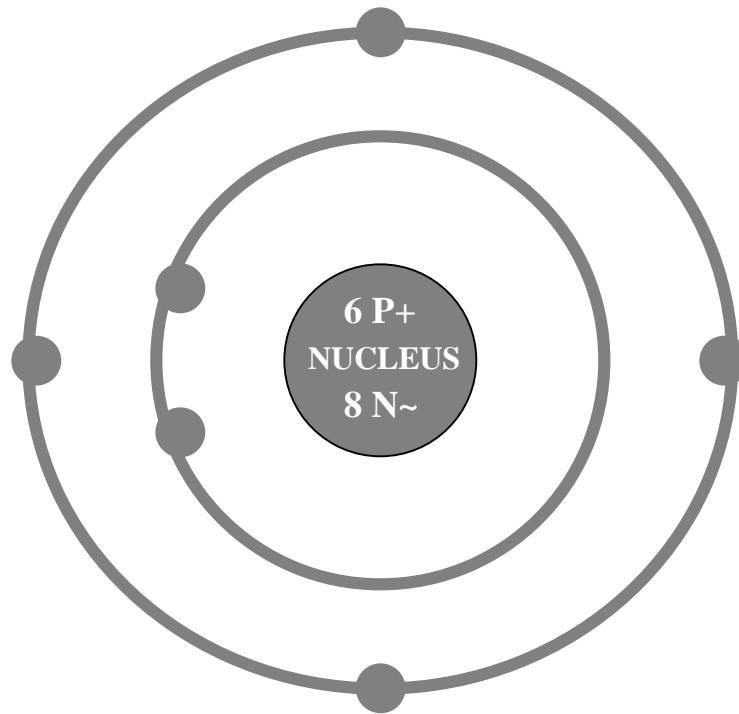


ATYPICAL CARBON ATOM





ATOMIC WEIGHT



CARBON AT. WT. = 14 DALTONS

● = e⁻



ATOMIC WEIGHT

A Bohr model of an atom is centered on the slide. It features a central grey sphere labeled "NUCLEUS" with several smaller grey spheres representing protons and neutrons. Surrounding the nucleus are several concentric red circles representing electron shells. Small grey spheres representing electrons are placed at various points on these shells. The text "ATOMIC WT. MAY VARY /ATOM/ELEMENT" is overlaid on the diagram in a large, white, serif font.

**ATOMIC WT.
MAY VARY
/ATOM/ELEMENT**

ATOMIC WEIGHT

QUESTION

WHAT DO SCIENTISTS
CALL AN ATOM
WITH AN ATYPICAL
NEUTRON NUMBER?

QUESTION



ANSWER

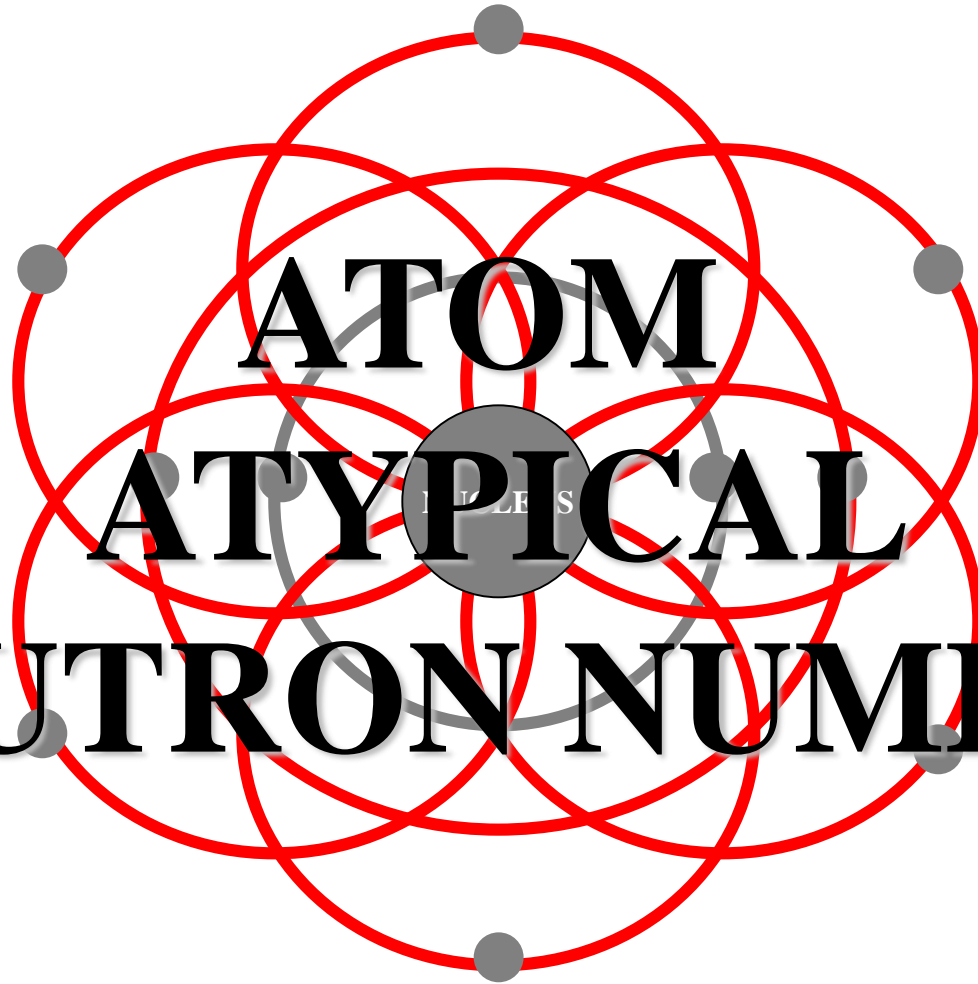
ISOTOPE

ANSWER

ISOTOPE

ISOTOPE

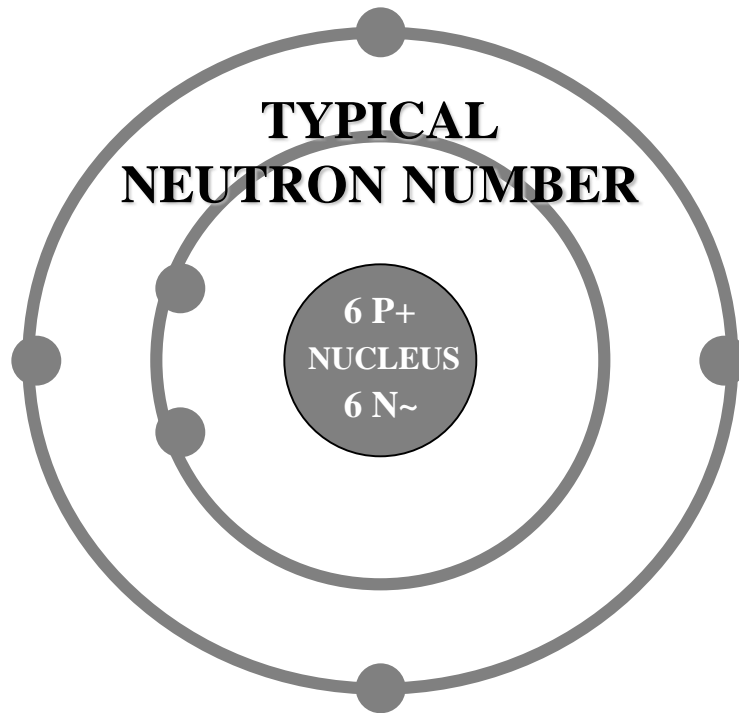
ISOTOPE



NEUTRON NUMBER

ISOTOPE

TYPICAL CARBON ATOM



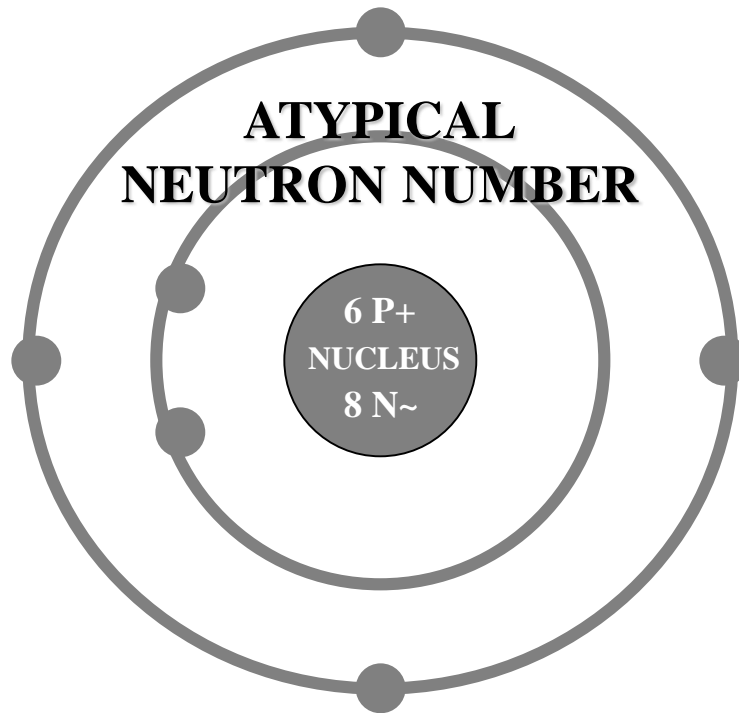
ATOMIC WT. = 12 DALTONS

C12

● = e-



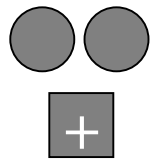
ISOTOPE CARBON ATOM



ATOMIC WT. = 14 DALTONS

C14

● = e-



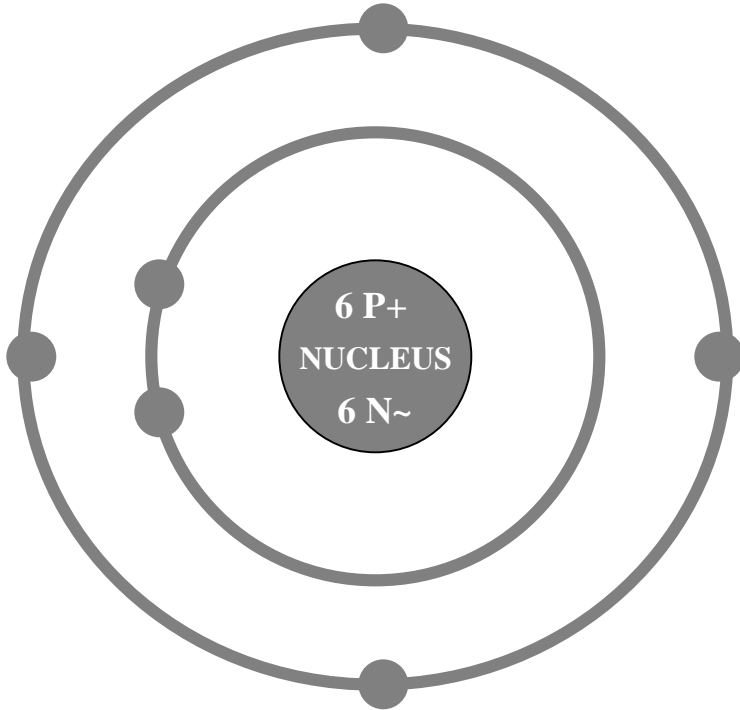
QUESTION

DO ISOTOPES OF
THE SAME ELEMENT
HAVE DIFFERENT
CHEMICAL PROPERTIES?

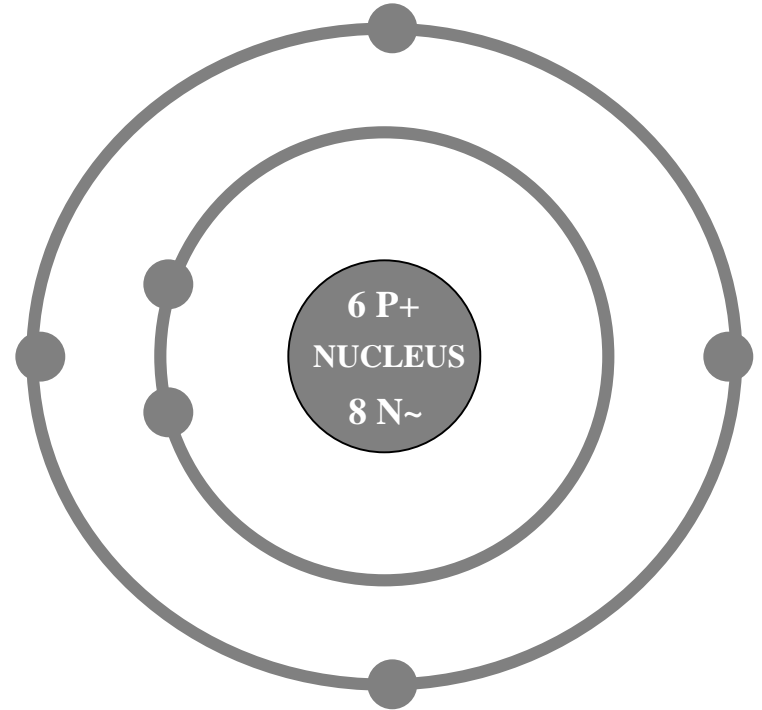
QUESTION



CARBON ATOMS



C12
TYPICAL



C14
ISOTOPE

● = E-

QUESTION

DO ISOTOPES OF
THE SAME ELEMENT
HAVE DIFFERENT
CHEMICAL PROPERTIES?

QUESTION



ANSWER

NO

ANSWER

CARBON ATOMS

ISOTOPE

CHEMICAL

PROPERTIES

DO NOT CHANGE



C12
TYPICAL

C12 & C14
CHEMICALLY
ACT AS
CARBON ATOMS

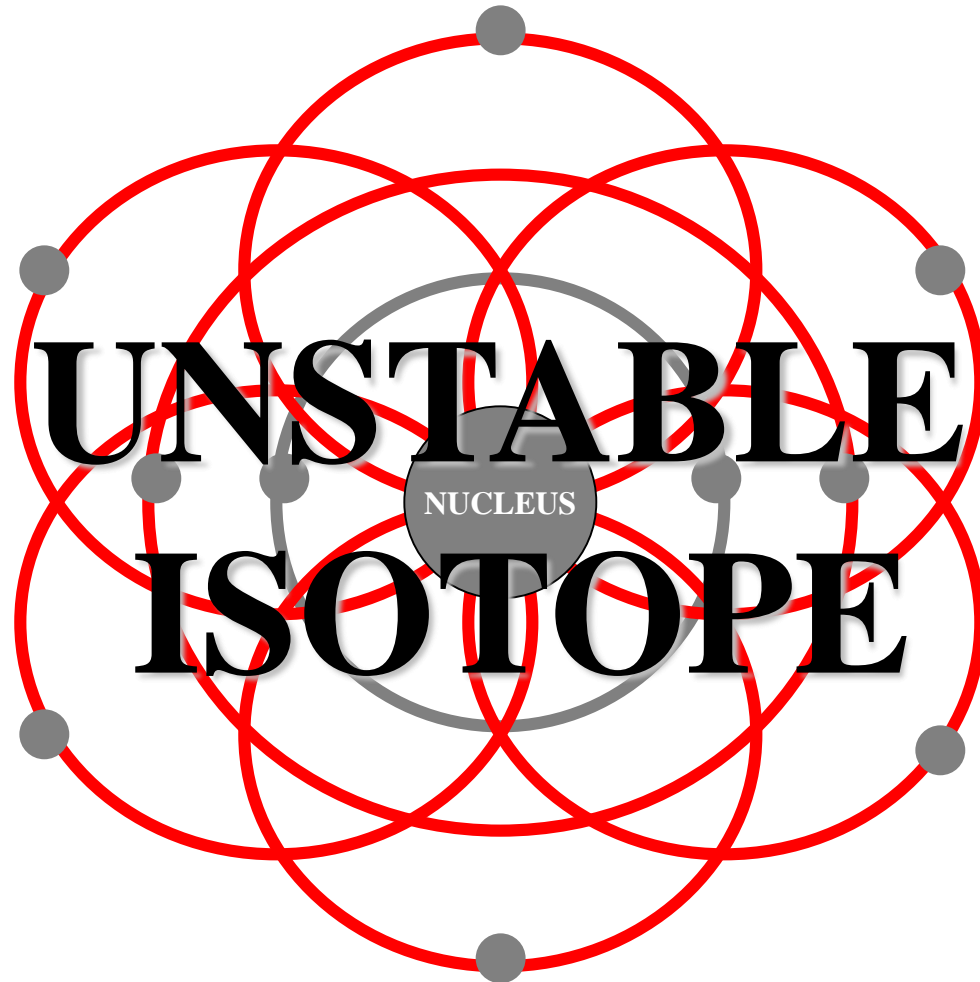
C14
ISOTOPE

● = e-

RADIOACTIVE ISOTOPE

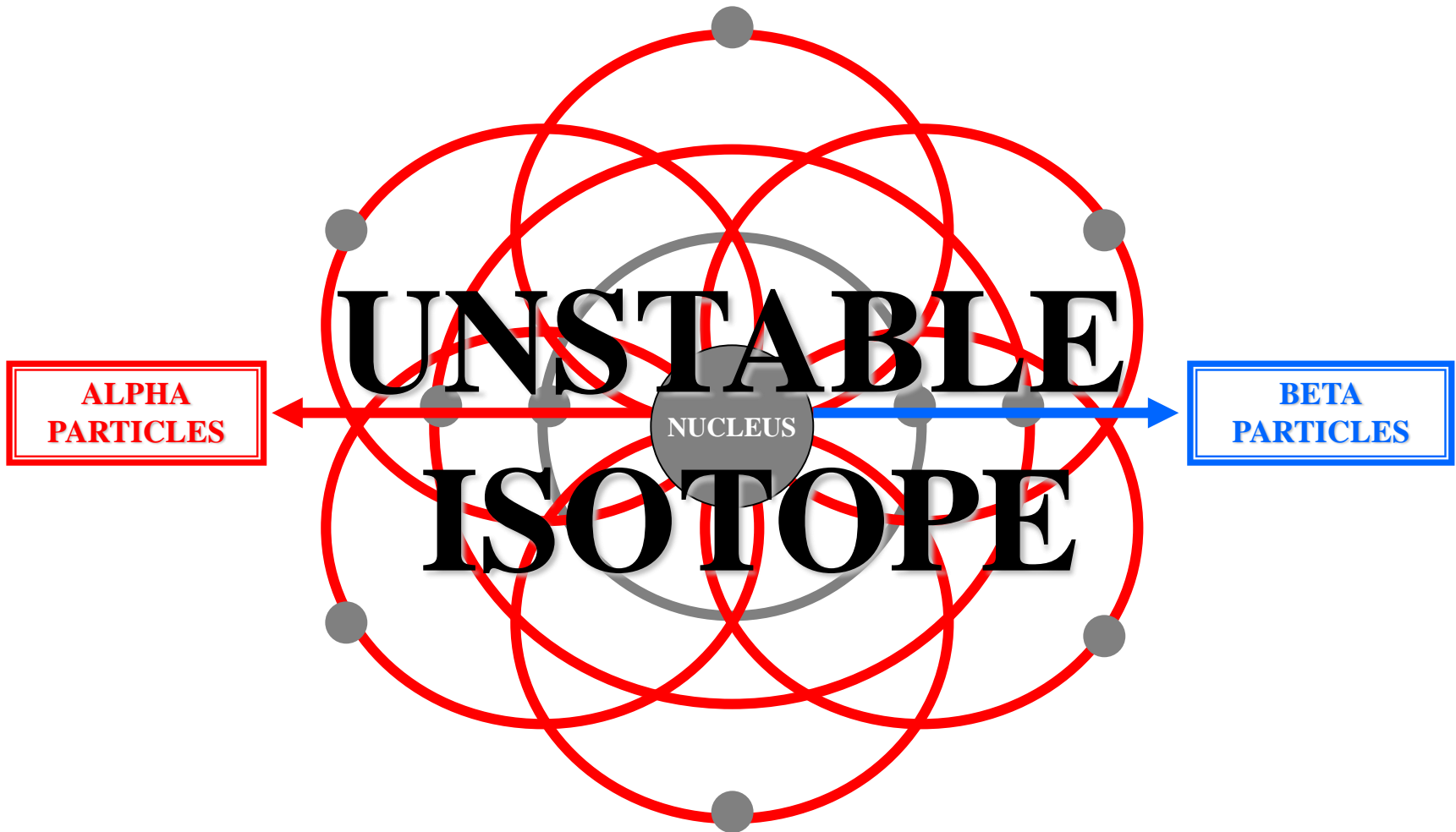
RADIOACTIVE ISOTOPE

RADIOACTIVE ISOTOPE



RADIOACTIVE ISOTOPE

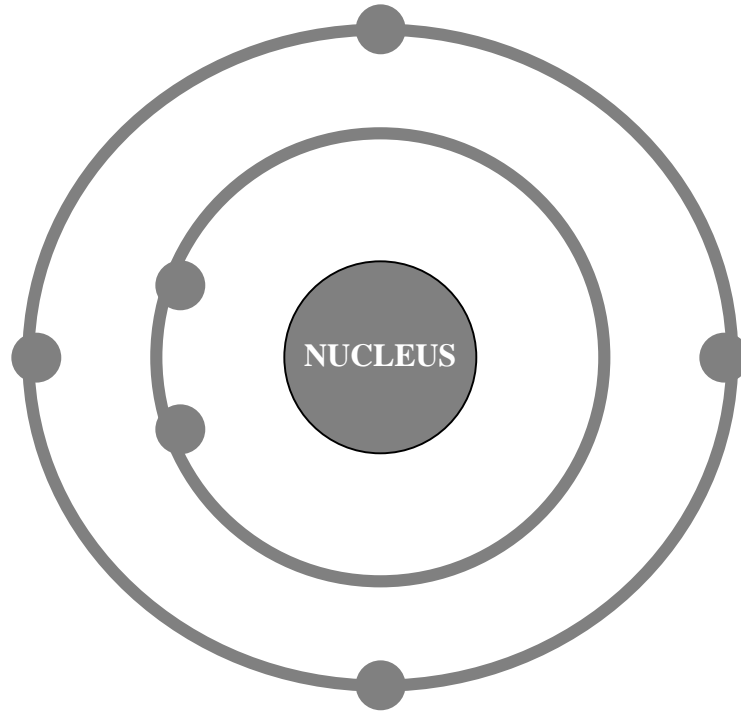
RADIOACTIVE ISOTOPE



EMITS **ALPHA** OR **BETA** PARTICLES

RADIOACTIVE ISOTOPE

UNSTABLE ISOTOPE

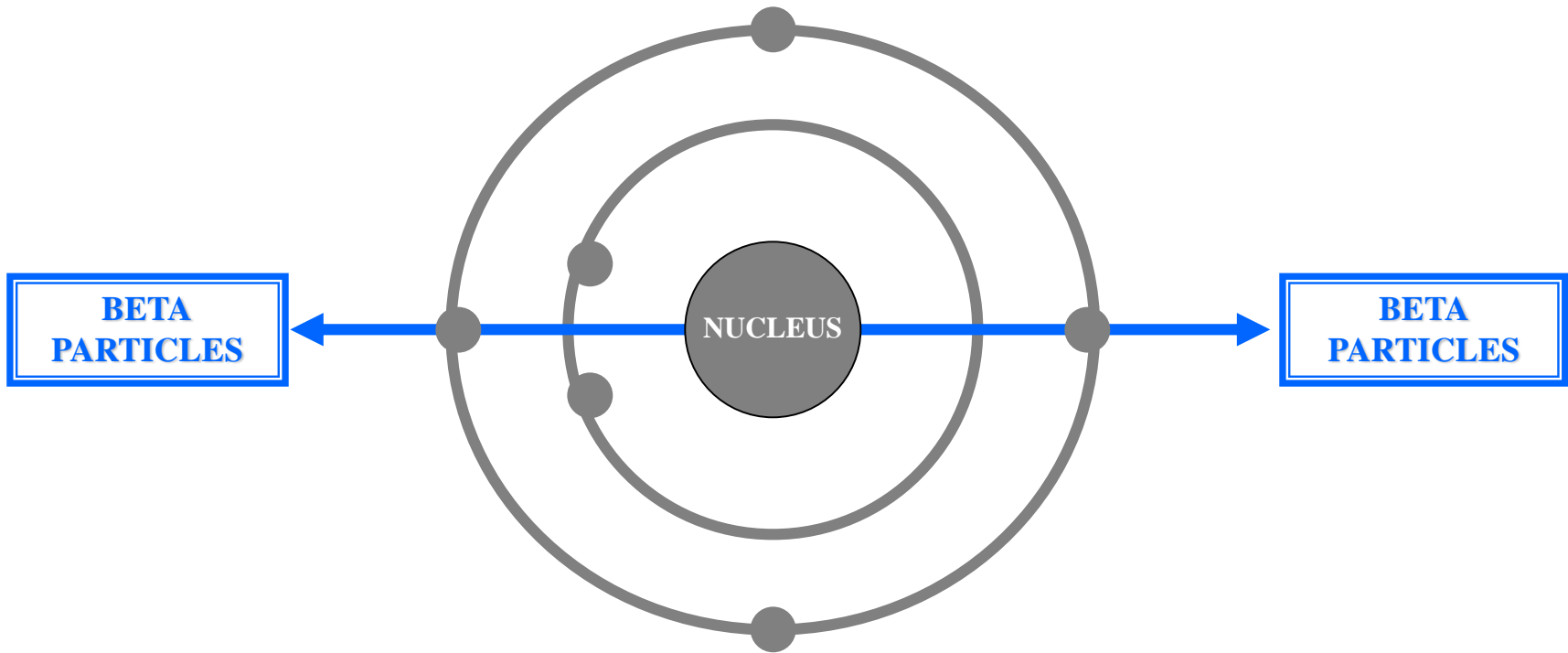


C14

● = e-

RADIOACTIVE ISOTOPE

UNSTABLE ISOTOPE



EMITS **BETA** PARTICLES

C14

● = e-



FD

C14
USEFUL
BIOLOGICAL
TOOL

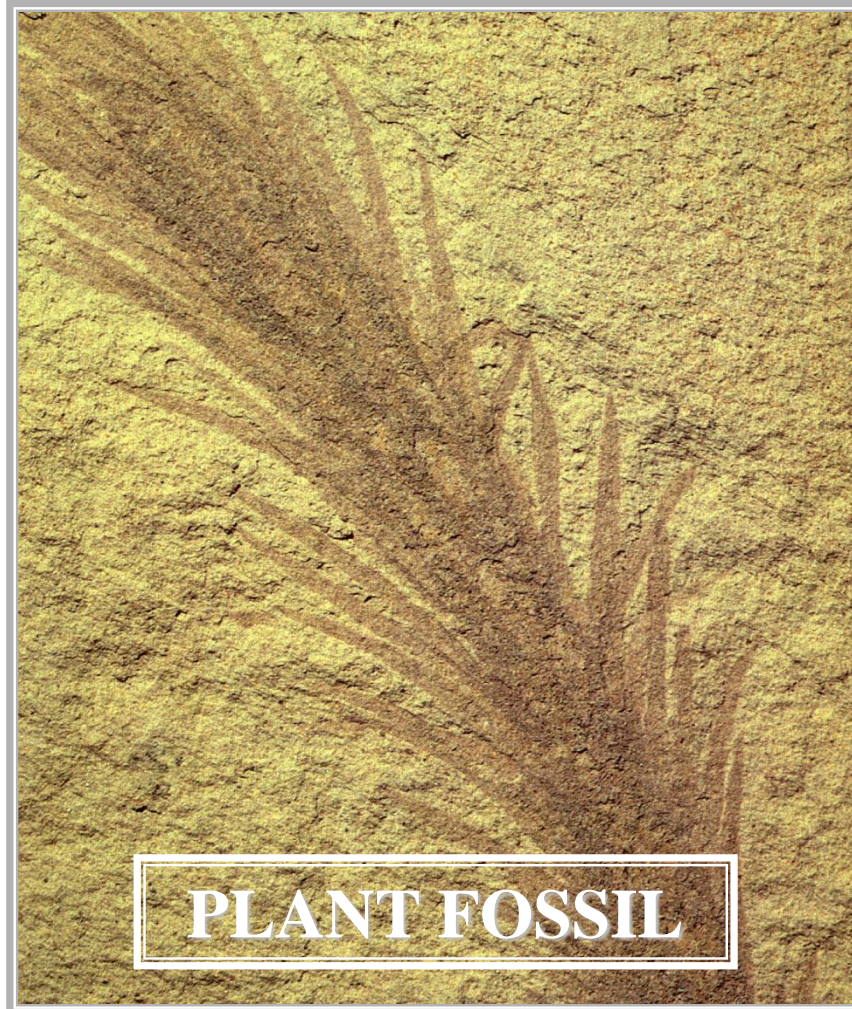


C14

FOSSIL DATING

C14: FOSSIL DATING

Radio-Carbon Dating



C14

BIOCHEMICAL

TRACER

C14

**BIOCHEMICAL
TRACER**

**USED TO ELICITATE
BIOCHEMICAL PATHWAYS**